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Development and Verification of Tritium Analyses Code for a Very High Temperature Reactor

Chang H. Oh Eung S. Kim September 2009



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Development and Verification of Tritium Analyses Code for a Very High Temperature Reactor

Chang H. Oh Eung S. Kim

September 2009

Idaho National Laboratory Next Generation Nuclear Plant Project Idaho Falls, Idaho 83415

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Next Generation Nuclear Plant Project

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Approved by:

Chang Oh Principal Investigator

Mike Patterson Engineering Technical Manager

Vince Tonc NGNP Engineering Director

 $2 \sim 0^{\circ}$ Date

Date

9/29/09

ABSTRACT

A tritium permeation analyses code (TPAC) was developed by Idaho National Laboratory for the purpose of analyzing tritium distributions in Very High Temperature Reactor (VHTR) VHTR systems, including integrated hydrogen production systems. A MATLAB SIMULINK software package was used for development of the code. The TPAC is based on the mass balance equations of tritium-containing species and various forms of hydrogen (i.e., HT, H₂, HTO, HTSO₄, and TI) coupled with a variety of tritium sources, sinks, and permeation models. In the TPAC, ternary fission and neutron reactions with ⁶Li, ⁷Li ¹⁰B, and ³He were taken into considerations as tritium sources. Purification and leakage models were implemented as main tritium sinks. Permeation of HT and H₂ through pipes, vessels, and heat exchangers were considered as the main tritium transport paths. In addition, electrolyzer and isotope exchange models were developed for analyzing hydrogen production systems, including hightemperature electrolysis and sulfur-iodine processes.

Tritium behaviors in a very high temperature reactor/high temperature steam electrolysis system were analyzed by the TPAC based on the reference indirect parallel configuration proposed by Oh et al. (2007). This analysis showed that only 0.4% of tritium released from the core is transferred to the product hydrogen. The amount of tritium in the product hydrogen was estimated to be approximately an order of magnitude less than the gaseous effluent limit for tritium.

The TPAC has unlimited flexibility for system configurations and provides easy drag-and-drops for making models by adopting a graphical user interface. Verification of the code was performed by comparisons with analytical solutions and the experimental data from the Peach Bottom reactor design. Calculations from the THYTAN code developed in Japan and adopted by Japan Atomic Energy Agency were also compared with the TPAC solutions. This report contains descriptions of the basic tritium pathways, theory, a simple user guide, verifications, sensitivity studies, sample cases, and code tutorials.

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Computer Platform Information	

ACRONYMS

ASME	American Society of Mechanical Engineers
AVR	Arbeitsgemeinschaft Versuchsreaktor (Germany)
BDF	backward differentiation formula
Bq	becquerel, the SI unit of radioactivity
CFD	computational fluid dynamics
Ci	curie
EPA	Environmental Protection Agency
FEM	Finite Element Method
GCI	grid convergence index
GUI	graphical user interface
HT	tritium gas
НТО	tritiated water
HTE	high-temperature electrolysis
HTR-10	High Temperature Reactor (10 MW) in China
HTGR	high temperature gas-cooled reactor
HTSE	high temperature steam electrolysis
HTTR	high temperature test reactor
HTU	heat transfer unit
IHTL	intermediate heat transfer loop
IHX	intermediate heat exchanger
LMTD	log mean temperature difference
NDF	numerical differentiation formula
NGNP	Next Generation Nuclear Plant
NRC	Nuclear Regulatory Commission
PCU	power conversion unit
PCHE	printed circuit heat exchanger
PHX	process heat exchanger
SHX	secondary heat exchanger
SI	sulfur-iodine

- TPAC Tritium Permeation Analyses Code
- VHTR very high temperature gas-cooled reactor

Development and Verification of Tritium Analyses Code for a Very High Temperature Reactor

1. INTRODUCTION

The very high temperature gas-cooled reactor (VHTR) is a helium-cooled, graphite moderated, thermal neutron spectrum nuclear reactor that will be operated at up to 9 MPa and 950°C. The VHTR is one of six reactor concepts recommended by the *Generation IV Technology Roadmap* for further development (DOE 2002). This reactor is currently envisioned as co-generation of electricity and process heat to many industries. It is regarded as one of the most promising future energy technologies because of its high efficiency, inherent passive safety features, and high temperature applications, including hydrogen production without CO_2 emissions to atmosphere. Although a final reference configuration has not been established, the NGNP is expected to operate at 750-800°C and 5 -9 MPa. The work presented here will be applicable to the NGNP Project as well as subsequent VHTR designs at higher temperatures and pressures. While specific configurations and components are assumed throughout this document, it should be understood that those assumptions are not final. Component and configuration changes – for example, use of steam generators instead of heat exchangers – may continue through the design process.

One potential problem of using VHTRs for hydrogen production applications is tritium permeation from the primary coolant to the hydrogen product through heat transfer surfaces. In VHTRs, tritium mostly comes from ternary fission of the fuel and neutron reactions with lithium impurities in the graphite, boron control materials in the reflector block, and ³He naturally occurring in the helium coolant. Tritium that enters or is formed in the helium primary coolant will be circulated or permeated to the secondary coolant and the intermediate heat transfer loop. The permeated tritium enters the product hydrogen through heat exchanger surfaces. The mechanisms of tritium transport are diffusion, bulk transport, and permeation (See Figure 1-1).



Figure 1-1. Diffusion, bulk transport, and permeation pathways of tritium in a notional heat transport loop system (Sherman and Adams 2008).

Tritium is a radioactive isotope of hydrogen with the half life of 12.32 years, and mainly generated in the reactor core by ternary fissions and neutron reactions (Compare et al. 1974). Table 1-1 summarizes the radioactive properties of tritium. The nucleus of a tritium atom consists of a proton and two neutrons. This contrasts with the nucleus of an ordinary hydrogen atom and a deuterium atom. Ordinary hydrogen comprises over 99.9%, deuterium comprises 0.02%, and tritium comprises about a 10^{-16} % of naturally occurring hydrogen. The chemical properties of tritium are essentially the same as ordinary hydrogen. Typically, tritium exists as a form of HT, tritium gas (¹H-³H), because of isotope exchange reactions between T₂ (³H-³H) and H₂ (Wichner and Dyer 1979). Tritiated water, HTO, is another common form of tritium. In tritiated water, a tritium atom replaces one of the hydrogen atoms so the chemical form is HTO rather than H₂O.

					Radiation Energy (MeV)		
	Half Life	Natural	Specific	Decay	Alpha	Beta	Gamma
Isotope	(yr)	Abundance	Activity	Mode	(α)	(β)	(γ)
Н-3	12	<<1	9,800	Beta (β)	_	0.0057	—

Table 1-1. Radioactive Properties of Tritium.

The effect of radiation from tritium on health is relatively small and hazardous only if it is taken into the body because tritium decays by emitting a low-energy beta particle with no gamma radiation. This beta particle cannot penetrate deeply into tissue or travel far in air. The most likely form of uptake is as tritiated water. Uptake of tritium gas is typically very low (less than 1%). The health hazard of tritium is associated with cell damages caused by the ionizing radiation that results from radioactive decay, with the potential for subsequent cancer induction. Nearly all (up to 99%) inhaled tritium can be taken into the body from the lungs, where circulating blood then distributes it to all tissues. Ingested tritium oxide is almost completely absorbed, moving quickly from the gastrointestinal tract to the bloodstream. Within minutes it is found in varying concentrations in body fluids, organs, and other tissues. Generally, tritium is uniformly distributed through all biological fluids within one to two hours. Tritium is eliminated from the body with a biological half-life of 10 days, the same as for water.

In the U.S., tritium emissions are regulated by the U.S. Nuclear Regulatory Commission (NRC) and U.S. Environmental Protection Agency (EPA). Table 1-2 summaries the regulatory constraints on tritium.

	Annual			Effluent Concentration			
		Radiation Dose		Air		Water	
	Regulation	(mrem)	(mSv)	(µCi/ml)	(Bq/ml)	(µCi/ml)	(Bq/ml)
Limit	10 CFR 20.1301(a) ^a	100	1	—		_	_
	Table 2 of Appendix B to 10 CFR 20	50	0.5	1 × 10 ⁻⁷	3.7×10^{-3}	1 × 10 ⁻³	37
Standard	10 CFR 20.1301(e)	25	0.25	$(5 \times 10^{-8})^{a}$	$(1.85 \times 10^{-3})^{a}$	$(5 \times 10^{-4})^{a}$	$(18.5)^{a}$
ALARA	Appendix I to	15	0.15	$(3 \times 10^{-8})^{a}$	$(1.11 \times 10^{-3})^{a}$	-	-
	10 CFR 50	3	0.03			$(6 \times 10^{-5})^{a}$	$(2.22)^{a}$
Drinking Water	EPA standard	4	0.04		—	2×10^{-5}	0.74

Table 1-2. Regulatory constraints on tritium in the U.S. (Ohashi and Sherman 2007).

a. Calculated by assuming the linear relationship between the annual dose of 50 mrem and the values in Table 2 of Appendix B of 10 CFR 20.

ALARA = as low as reasonably achievable

CFR = Code of Federal Regulations	
EPA = U.S. Environmental Protection Agency	

Because of its small size, the tritium atom can easily penetrate through metal by diffusion. Especially in VHTRs, large portions of tritium in the primary coolant permeate through the heat exchanger surfaces of the intermediate heat exchanger (IHX) and process heat exchanger (PHX). It then mixes into the process chemicals of the hydrogen plant eventually contaminating the hydrogen products (Ohashi and Sherman 2007).

Tritium behavior in high temperature gas-cooled reactors (HTGR) was evaluated in several countries in the 1970s, e.g., the Dragon reactor in England (Forsyth 1974), the Peach Bottom HTGR in the U.S. (Wichner and Dyer 1979), and Arbeitsgemeinschaft Versuchsreaktor (AVR) in Germany (Steinwarz, Rohrig, and Nieder 1980). Data from the operation of HTGRs and from laboratory experiments revealed the mechanism of tritium production, transport, and release to the environment. In his review of tritium behavior in an HTGR system, Gainey's (1976) calculations show that tritium releases should be well within current federal guidelines for the nuclear plant. For example, the estimated maximum dose to an average adult for a typical 3,000-MWt HTGR with a cooling tower is 0.38 mrem/year, which is slightly more than one-tenth of the maximum annual dose allowed (Gainey 1976). For this reason, no further laboratory-scale work on tritium release and did not examine questions related to nuclear hydrogen production. NGNP will use the HTGR as a heat source to produce hydrogen for industry or individual users.

Ohashi and Sherman (2007) recently estimated steady-state tritium movement and accumulation in an NGNP with a hydrogen plant using a high temperature electrolysis process and a thermochemical water splitting sulfur-iodine (SI) process using the THYTAN code. Estimated tritium concentrations in product hydrogen and in process chemicals in the hydrogen plant of the NGNP using the high temperature electrolysis process were slightly higher than the drinking water limit defined by the EPA and the limit in the effluent at the boundary of an unrestricted area of a nuclear plant as defined by the NRC. However, modified designs and operations could bring these concentrations within specified limits. Tritium concentrations in the NGNP using the SI process were calculated to be significantly higher and were affected by parameters with large uncertainties (tritium permeability of the PHX, the hydrogen concentration in the heat transfer and process fluids, and the equilibrium constant of the isotope exchange reaction between HT and H₂SO₄). These parameters, including tritium generation and the release rate in the reactor core, should be more accurately estimated in the near future to improve the calculations for the SI process. Decreasing the tritium permeation through the heat exchanger between the primary and secondary circuits by means of tritium permeation barriers, capture systems, and mitigation measures may be an effective method for decreasing tritium concentrations in product hydrogen, hydrogen plant, and tertiary coolant.

The main objective of the current research is to develop a computer code to analyze tritium behaviors in the VHTR systems as continuation of FY-08 work. The code is named "Tritium Permeation Analyses Code" (TPAC), and it is written in MATLAB SIMULINK Package. The TPAC was originally developed for analyses of tritium production and distribution in VHTRs and hydrogen production systems. However, the capability is not limited to only VHTR systems. Flexibility of the code can be extended by simple modifications to the general chemical systems that include tritium. The TPAC can solve the generations of tritium source and tritium transport equations, including the penetration of tritium through the heat exchanger wall. The details are described in this report.

2. SOURCES AND PATHWAYS OF TRITIUM IN VHTR

2.1 Tritium Source

The tritium generation mechanism in VHTRs is well described by Gainey (1976), and Ohashi and Sherman (2007). The primary tritium birth mechanism is ternary fission of fuel (e.g., ²³³U, ²³⁵U, ²³⁹Pu, and ²⁴¹Pu) because of thermal neutrons. Tritium is also generated in VHTR from ⁶Li, ⁷Li, ³He, and ¹⁰B by neutron capture reactions as:

${}^{6}\text{Li}(n, \alpha) {}^{3}\text{H},$	(2-1)
7_{1} : (m, max) 3_{11}	(2, 2)

$$\frac{1}{2} L_{1}(n, n\alpha)^{2} H$$

$$(2-2)$$

3
He (n, p) 3 H (2-3)

$$^{10}B(n, 2\alpha)^{3}H$$
 (2-4)

 10 B (n, α) ⁷Li. (2-5)

⁶Li and ⁷Li are impurities in the core graphite material such as the sleeve, spine, reflector, and fuel matrix. ³He is an impurity in the reactor coolant helium. Because helium coolant leaks from the primary loop to the containment vessel, helium is supplied to the primary coolant as a make-up with an impurity of ³He. ¹⁰B exists in control rods, burnable poisons, and reflectors. Tritium is produced directly from ¹⁰B via Equation (2-4) or via the chain reaction in Equations (2-2) and (2-5).

2.2 Distribution of Tritium

Tritium generated in the fuel particles by ternary fissions can escape into a primary coolant, permeating several barriers of the fuel particles. In addition, tritium born from ¹⁰B and ⁶Li can pass into the primary coolant. The principal chemical form for tritium in the reactor coolant was reported as HT (¹H-³H) because of the isotope exchange reaction between T_2 (³H-³H) and H_2 (Wichner and Dyer 1979). Some of the tritium in the primary coolant is removed by a purification system installed in the primary loop. Some of the tritium can escape outside the coolant by permeation through the components and piping and by leakage with the primary helium coolant. The remaining tritium in the primary coolant permeates through the heat transfer tubes or surfaces of the IHX and gets mixed in with the secondary coolant.

In the secondary loop, some of the tritium is removed by the purification system or escapes outside, just as tritium behaves in the primary loop. The remainder of the tritium in the secondary coolant permeates through heat transfer surfaces and gets mixed into the tertiary coolant. Transportation of tritium into the tertiary coolant is the same as for the secondary coolant. It permeates through the heat transfer surfaces of PHXs and is mixed into the process chemicals of the hydrogen plant.

Tritium that has permeated the tertiary loop going to the hydrogen plant can react with hydrogen-containing process chemicals through isotopic exchange reactions. For example, the high-temperature electrolysis (HTE) process contains H_2O . Therefore, HTO (tritiated water) is produced by the isotope exchange reaction between HT and H_2O . Gaseous HTO and HT escape from the HTE process with the product hydrogen and oxygen. A portion of the liquid HTO flows out from the hydrogen plant with the drain water. The remainder of the liquid HTO is circulated into the plant with the recycling water and is accumulated in the water of the HTE process. The SI process used in hydrogen production also contains H_2O , H_2SO_4 , and HI chemicals from which HTO, HTSO₄, and TI may be produced through isotope exchange reactions. These tritium-containing chemicals are circulated with and are accumulated

in process the chemicals. Only gaseous HT and HTO can escape from the SI process with the product hydrogen and oxygen.

The above tritium pathways in VHTRs are illustrated in Figure 2-1. All of the production mechanisms lead to either retention in solids or entry into coolant systems. The tritium in the primary coolant can be adsorbed on graphite surfaces, leaked out of the primary circuit, removed from the purification system, or permeated to the process heat application system like hydrogen production.



Figure 2-1. General tritium pathways in VHTRs.

Figure 2-2 shows the tritium distribution calculated by Ohashi and Sherman (2007) for a VHTR/HTE system at steady state. The calculation was performed with the THYTAN code developed in Japan and used by the Japan Atomic Energy Agency (JAEA) (Ohashi and Sherman 2007). According to their calculations, 30.7% of the tritium released from the core is removed by the purification system in the primary loop. A small portion of tritium (~0.11%) leaks to the outside from the primary loop with helium leakage. The remainder of the tritium (69.1%) permeates into the secondary loop through a heat exchanger. For the secondary loop, 30.3% of the tritium, which is about one half of the permeated tritium from the primary loop, is removed by the purification system. Consequently, 38.7% of tritium permeates from the secondary loop is removed by the purification system. Tritium permeates all of the tritium permeated from the secondary loop is removed by the purification system. Tritium permeation rates from the tertiary loop to the main line and to the sweep line of the HTE process are 0.946% and 2.499%, respectively. However, depending on the system designs and operating conditions, the specified values can be completely different from their base calculations.



Figure 2-2. Tritium distribution in the VHTR system (Ohashi and Sherman 2007).

3. THEORY

This section summarizes the theory adopted in the TPAC. The TPAC is capable of modeling tritium behaviors in the simplified representations of the VHTR, HTE, and SI processes. The following phenomena are taken into consideration in this code:

- Mass balance of the species (H₂, HT, HTO, HTSO₄, and TI)
- Tritium and hydrogen permeation through the heat transfer surface
- Tritium and hydrogen permeation to outside through the pipe wall
- Tritium and hydrogen removal by the purification system
- Tritium and hydrogen leakage by helium leakage
- Isotope exchange between tritium containing and hydrogen containing chemicals
- Species movement in the electrolyzer.

3.1 Mass Balance

The TPAC basically solves the mass balance of tritium-containing chemicals and hydrogen in each component block as follows:

$$V_{j} \frac{dC_{i,j}}{dt} = \sum \left(F_{total,j-1} \cdot C_{i,j-1} \right) - F_{total,j} \cdot C_{i,j} - S_{i,j}$$
(2-6)

$$S_{i,j} = -R_{core,i,j} + R_{HX,i,j} + R_{copipe,i,j} + R_{comp,i,j} + R_{leak,i,j} + R_{PF,i,j} + R_{reaction,i,j}$$
(2-7)

where

V_{j}	=	the volume of block j [m3 (STP)]
$C_{i,j}$	=	the volume fraction of chemical i in block j [m3 (STP)/m3 (STP)]
t	=	time [s]
$F_{total,j}$	=	the volumetric flow rate of all chemicals in block j [m3 (STP)/s]
$S_{i,j}$	=	the total amount of volume change rate of chemical i in block j by considering generation, release, permeation, removal, leakage, and isotope exchange reactions [m3 (STP)/s]
$R_{core,i,j}$	=	volumetric release rate from the core to the primary coolant [m3 (STP)/s]
$R_{HX,i,j}$	=	volumetric permeation rate at heat exchanger ($i = H2$ and HT) [m3 (STP)/s]
$R_{copipe,i,j}$	=	volumetric permeation rate at co-axial pipe (i = H2 and HT) [m3 (STP)/s]
$R_{comp,i,j}$	=	volumetric permeation rate to outside ($i = H2$ and HT) [m3 (STP)/s]
$R_{leak,i,j}$	=	volumetric leak rate with helium leakage ($i = H2$, HT and HTO) [m3 (STP)/s]
$R_{PF,i,j}$	=	volumetric removal rate by purification system (i = H2, HT and HTO) $[m3 (STP)/s]$

 $R_{reaction,i,j}$ = volumetric reaction rate by isotope exchange reactions (i = H2, HT, HTO, HTSO4 and TI) [m3 (STP)/s].

3.2 Tritium Source

There are five different sources for tritium generation in the reactor core: ternary fission, birth from ⁶Li, birth from ⁷Li, birth from ³He, and birth from ¹⁰B. Table 3-1 summarizes the tritium production sources and total contributions. The detail equations for each source are described in the sections that follow.

Tritium Source	% total	Release Potential
Ternary Fission	62	Time at temperature
From ³ He	18	Produced in He coolant
From ⁶ Li		Partly retained in graphite;
• Core Graphite	2	Release during H ₂ O ingress
Core Matrix	10	
• Reflector (replaceable)	<1	
• Reflector (permanent)	<1	
From ¹⁰ B		Apparently retained at Source
Control Rod	7	
Burnable Poisson	1	
• Reflector	<1	

Table 3-1. Sources of tritium production in VHTRs.

3.2.1 Ternary Fission

Tritium yields formed by ternary fission depend on the uranium isotope undergoing fission and the neutron flux.

$$\frac{d(N_{T(Ter)})}{dt} = K \cdot P \cdot Y - \lambda \cdot N_{T(Ter)}$$
(3-1)

where

 $N_{T(Ter)}$ = number of tritium atoms from ternary fission

K = fission rate per thermal megawatt [fission/MW/s]

P = reactor power [MW]

 λ = tritium decay constant [1/s].

Table 3-2 summarizes the necessary information for calculating tritium production by ternary fission.

	Unit	Values
Reactor Power (P)	MWth	600
Fission Rate (K)	Fissions/MW sec	3.12e16
Average Yield Per Fission (Y)	1/fission	1.0e-4
Failure Fraction of Fuel Particles	-	0.3

Table 3-2. Basic data for THYTAN evaluation of ternary fission.

3.2.2 Birth from ⁶Li

As described, tritium is produced in the VHTR by various sources such as ternary fissions and activation reactions with impurities and boron in the materials. Table 2-13 summarizes the tritium production reactions and cross sections (Wichner and Dyer 1979).

Table 3-3. Tritium production reaction and cross section (Wichner and Dyer 1979).

Production Reaction	Cross section (barns)	Energy range (eV)	
³ He(n,p)T	2,280	0-2.38	
⁶ Li(n,a)T	408	0-2.38	
$^{10}B(n,2a)T$	0.050	>0.18×10 ⁶	
${}^{10}B(n,a)^{7}Li$	1,630	0-2.38	
⁷ Li(n,n'a)T	0.153	>0.18×10 ⁶	

The birth of tritium from ⁶Li, which is an impurity in the core and reflector graphite materials, is estimated by

$$\frac{d(N_{Li6})}{dt} = -\phi_{th} \cdot \sigma_{Li6T} \cdot N_{Li6} \text{ and}$$

$$\frac{d(N_{T(Li6)})}{dt} = \phi_{th} \cdot \sigma_{Li6T} \cdot N_{Li6} - \lambda \cdot N_{T(Li6)}$$
(3-2)
(3-3)

where

 N_{Li6} = number of ⁶Li atoms

 $N_{T(Li6)}$ = number of tritium atoms from ⁶Li

$$\phi_{th}$$
 = thermal neutron flux [neutrons/cm²/s]

 σ_{Li6T} = effective cross section for ⁶Li (n, α) ³H [cm²].

3.2.3 Birth from ⁷Li

The birth of tritium from ⁷Li is estimated by

$$\frac{d(N_{Li7})}{dt} = -\phi_f \cdot \sigma_{Li7H3} \cdot N_{Li7} \text{ and}$$
(3-4)

$$\frac{d(N_{T(Li7)})}{dt} = \phi_f \cdot \sigma_{Li7T} \cdot N_{Li7} - \lambda \cdot N_{T(Li7)}$$
(3-5)

where

 N_{Li7} = number of ⁷Li atoms, excluding ¹⁰B source

 $N_{T(Li7)}$ = number of tritium atoms from ⁷Li, excluding birth from ¹⁰B

 ϕ_f = fast neutron flux [neutrons/cm²/s]

 σ_{Li7T} = effective cross section for ⁷Li (n, n\alpha) ³H [cm²].

3.2.4 Birth from ³He

Tritium is produced from ³He via n (n,p) reaction with thermal neutrons in the primary coolant. The natural abundance of ³He in helium ranges from 1.3×10^{-6} to 2×10^{-7} . The ³He reaction has a relatively high thermal neutron cross section. The following equations express tritium birth from ³He.

$$\frac{d(N_{He3})}{dt} = f \cdot N_{He3}^{\circ} - f \cdot N_{He3} - \phi_{He} \cdot \sigma_{He3T} \cdot N_{He3}$$
(3-6)

$$\frac{d(N_{T(He3)})}{dt} = \phi_{He} \cdot \sigma_{He3T} \cdot N_{He3} - \lambda \cdot N_{T(He3)}$$
(3-7)

$$\phi_{He} = \frac{W_{core}}{W_{total}} \cdot \phi_{th} \tag{3-8}$$

where

,

 N_{He3} = number of ³He atoms

 $N_{T(He3)}$ = number of tritium atoms from ³He

- f = fractional supply rate of helium coolant [1/s]
- N_{He3}° = number of ³He atoms in the supply helium
- σ_{He3T} = effective cross section for ³He (n, p) T [cm²]

$$\phi_{He}$$
 = average thermal neutron flux experienced by the total primary helium inventory [n/cm²/s]

$$W_{core}$$
 = helium inventory in core [kg]

 W_{total} = total primary helium inventory [kg].

3.2.5 Birth from ¹⁰B

Tritium birth from 10B can be expressed by

$$\frac{d(N_{B10})}{dt} = -\left(\phi_{th} \cdot \sigma_{B10Li7} + \phi_f \cdot \sigma_{B10T}\right) \cdot N_{B10}$$
(3-9)

$$\frac{d(N_{Li7(B10)})}{dt} = \phi_{th} \cdot \sigma_{B10Li7} \cdot N_{B10} - \phi_f \cdot \sigma_{Li7T} \cdot N_{Li7(B10)}$$
(3-10)

$$\frac{d(N_{T(B10)})}{dt} = \phi_f \cdot \sigma_{Li7T} \cdot N_{Li7(B10)} + \phi_f \cdot \sigma_{B10T} \cdot N_{B10} - \lambda \cdot N_{T(B10)}$$
(3-11)

where

 $N_{B10} = \text{number of } {}^{10}\text{B atoms}$ $N_{Li7(B10)} = \text{number of } {}^{7}\text{Li atoms from } {}^{10}\text{B}$ $N_{T(B10)} = \text{number of tritium from } {}^{10}\text{B}$ $\sigma_{B10Li7} = \text{effective cross section for } {}^{10}\text{B (n, \alpha) } {}^{7}\text{Li [cm^2]}$ $\sigma_{B10T} = \text{effective cross section for } {}^{10}\text{B (n, 2\alpha) } {}^{3}\text{H [cm^2]}.$

3.2.6 Tritium Release Rate

After formation, tritium can rest in solids or circulate in the coolant. The retention of tritium in solids is well summarized in the GA report published in 2006 (GA-911081). The following description is the guideline for determining tritium retention fraction proposed by GA:

A steady-state retention fraction is assigned for each reaction chain. In the fuel, it is assumed that tritium formed in cracked particles or from contaminants on particles escapes readily while that formed within intact TRISO particles is strongly retained. Less than 1% of the particles are expected to be defective or fail during irradiation. Therefore, a tritium retention fraction of 0.99 is generally recommended. However, more investigation needs to be performed later.

The chains that involve reactions of nuclides in the solid phase to produce tritium depend on the content and location of possible boron and lithium impurities in graphite, which are not well known. The release rate of tritium, which is bound interstitially in graphite, is expected to be quite slow because a temperature of at least 1,200°C is required for tritium desorption from graphite. A retention fraction of 0.99 was recommended as an input to TRITGO (General Atomics 2006) for these reaction chains.

Each region solid is assumed to have uniform properties, but this is not necessarily a good assumption for reflector blocks containing localized boron pins. Treating these regions as if boron were uniformly dispersed throughout the graphite may cause the average reaction rate for this boron to be too high. In the case of control rods, the major lithium content is in the metal. The retention of tritium formed from this source may be over-predicted because tritium escapes more readily from metal than from graphite.

The control rods will not be fully inserted. An insertion fraction of about 11% is frequently cited. The relationship between extra gas space under rods, degree of rod insertion, etc., has not been treated as a variable.

Tritium from the 3 He(n,p)T reaction in the primary coolant will come to rest in solids as fractions, which vary from region to region depending on the space associated with channels and fabricated holes, clearance annuli, and pores. The following paragraphs summarize the method to calculate the average fraction of tritium that recoils into solids.

The recoil energy of a tritium (³H) is about 0.2 MeV, leading to an estimated range of 0.05 cm in helium at 47.6 atm and 1,000°C. In the case of pores, because they are generally smaller than 0.05 cm, the fraction bound is taken as unity. A planar approximation is used for the clearance annuli (tolerances between graphite blocks or between the fuel rods and the graphite blocks containing them). If the width of the gap exceeds the recoil range,

Fraction Bound = range /
$$(2 \cdot width)$$
, range < width, (3-11)

If the width of the gap is less than the recoil range,

$$Fraction Bound = (range - width/2) / range, range > width$$
(3-12)

For large cylindrical channels or holes,

Fraction Bound = range / channel diameter

Based on above guidelines, the tritium release rate from the core to the primary coolant, $R_{core,HT,j}$ [m³ (STP)/s], is calculated using the following equations:

(3-13)

$$\frac{N_{T(total)}}{dt} = \sum \left\{ \alpha_{Ter} \frac{d(N_{T(Ter)})}{dt} \right\} + \sum \left[\alpha_{Li} \left\{ \frac{d(N_{T(Li6)})}{dt} + \frac{d(N_{T(Li7)})}{dt} \right\} \right] + \sum \left\{ \alpha_{He3} \frac{d(N_{T(He3)})}{dt} \right\} + \sum \left\{ \alpha_{B10} \frac{d(N_{T(B10)})}{dt} \right\}$$
(3-14)

$$R_{core,HT,j} = \frac{N_{T(total)}}{dt} \times \frac{1}{N_A} \times \frac{R \cdot T_0}{P_0}$$
(3-15)

where

 $N_{T(release)}$ = number of tritium atoms released to the primary coolant

α_{Ter}	=	fractional release ratio of tritium produced from ternary fission
$\alpha_{{\scriptscriptstyle L}i}$	=	fractional release ratio of tritium produced from ^{6}Li and ^{7}Li
α_{He3}	=	fractional release ratio of tritium produced from ³ He
α_{B10}	=	fractional release ratio of tritium produced from ${}^{10}\mathrm{B}$
N_A	=	Avogadro constant
R	=	gas constant (8.314)
T_0	=	standard temperature (273.15 K)
P_0	=	standard pressure (1.0 1325×10^5 Pa).

3.3 Permeation

In the VHTR system, heat exchangers are the main route of tritium permeation. It is therefore very important to predict accurate tritium penetration rates when estimating the tritium distributions and contamination levels in hydrogen from the hydrogen production plant. The permeation rate of H₂ at the heat exchanger, $R_{HX,H2}$ [m³(STP)/s], and at the co-axial pipe, $R_{copipe,H2}$ [m³(STP)/s], is generally estimated using

$$R_{HX,H2,j}\left(or \quad R_{copipe,H2,j}\right) = \frac{A}{l} \cdot k_{p,H} \cdot \left(\sqrt{P_{H2,h}} - \sqrt{P_{H2,l}}\right)$$
(3-16)

where

A = heat transfer area or surface area [m²] l = thickness of heat transfer tube or component casing [m] $k_{p,H} = permeability of hydrogen [m³(STP)/m/s/Pa^{0.5}]$

 $P_{H2,h}$ = partial pressure of H₂ at high pressure side [Pa]

$$P_{H2,l}$$
 = partial pressure of H₂ at low pressure side [Pa]

l is calculated by

$$l = r_o \cdot \ln\left(\frac{r_o}{r_i}\right) \tag{3-17}$$

where

 r_o = outer radius of heat transfer tube [m]

 r_i = inner radius of heat transfer tube [m].

Permeability is calculated by the Arrhenius equation:

$$k_p = f \cdot \exp\left(\frac{-E}{RT}\right) \tag{3-18}$$

where

f = pre-exponential factor of permeability [m³ (STP)/m/s/Pa^{0.5}]

E = activation energy [J/mol]

- R = ideal gas constant [J/mol-K]
- T = absolute temperature [K].

The permeation rate of H₂ though the outer wall of the component and piping, $R_{comp,H2}$, is calculated by excluding the H₂ partial pressure at the low pressure side as

$$R_{comp,H2,j} = \frac{A}{l} \cdot k_{p,H} \cdot \sqrt{P_{H2,h}} \quad . \tag{3-19}$$

The permeation rate of HT at the heat exchanger, $R_{HX,H2}$ [m³ (STP)/s], and at the co-axial pipe, $R_{copipe,H2}$ [m³ (STP)/s], is estimated by considering the effect of the existence of hydrogen on the adsorption-dissociation and recombination-desorption step as

$$R_{HX,HT,j}\left(or \quad R_{copipe,HT,j}\right) = \frac{A}{l} \cdot k_{p,T} \cdot \left(1 - \frac{P_{H2,l}}{P_{Total,l}}\right) \left(\frac{P_{HT,h}}{\sqrt{P_{H2,h} + P_{HT,h}}} - \frac{P_{HT,l}}{\sqrt{P_{H2,l} + P_{HT,l}}}\right)$$
(3-20)

where

 $k_{p,T}$ = permeability of tritium [m³(STP)/m/s/Pa^{0.5}]

 $P_{HT,h}$ = partial pressure of HT at high pressure side [Pa]

 $P_{HT,l}$ = partial pressure of HT at low pressure side [Pa]

 $P_{Total,l}$ = total pressure at low pressure side [Pa].

The permeation rate of HT through the outer wall of the component and piping, $R_{comp,HT}$, is calculated by

$$R_{comp,HT,j} = \frac{A}{l} \cdot k_{p,T} \cdot \frac{P_{HT,h}}{\sqrt{P_{H2,h} + P_{HT,h}}}$$
(3-21)

As shown in Equations (3-16), (3-17), (3-20), and (3-21), a tube thickness is a very sensitive factor for the tritium permeation. The tritium permeation rate is inversely proportional to the tube thickness. In the typical tubular type heat exchangers, determination of the tube thickness is simple and straight forward. However, the determination is not simple in the printed circuit heat exchanger (PCHE), which is one of the strong candidates for VHTR heat exchangers. Figure 3-1 shows the cross-sectional configurations of a PCHE. A PCHE is a category of the compact heat exchanger manufactured by HeatricTM (2003). It is manufactured by chemical etching and a diffusion bonding process, so it provides a large heat transfer surface area density - as high as 2,500 m²/m³. It allows operation at high temperatures and pressures up to 900°C and 50 MPa and requires a correlation for metal thicknesses because the tritium permeation path in the PCHEs is two dimensional and the thickness varies along the channel surface.



Figure 3-1. Cross-sectional picture of a PCHE (Dewson and Grady 2003).

Oh and Kim (2008) proposed the following correlations to determine the effective thickness (t_{eff}) of the PCHE channels based on two different configurations: standard configuration and offset configuration (See Figure 3-2).

1. Standard Configuration:

$$\frac{t_{eff.}}{d} = 0.8 \cdot \left(\frac{t_p}{d}\right)^{2.92} \cdot \left(\frac{p}{d}\right)^{-0.19} \qquad \left(\frac{t_p}{d} < 0.8\right)$$
(3-22)

$$\frac{t_{eff.}}{d} = 0.67 \cdot \left(\frac{t_p}{d}\right)^{1.51} \cdot \left(\frac{p}{d}\right)^{-0.61} \qquad \left(\frac{t_p}{d} > 0.8\right)$$
(3-23)

2. Offset Configuration:

$$\frac{t_{eff.}}{d} = 0.48 \cdot \left(\frac{t_p}{d}\right)^{1.94} \cdot \left(\frac{p}{d}\right)^{-0.85} \qquad \left(\frac{t_p}{d} < 0.8\right)$$
(3-24)

$$\frac{t_{eff.}}{d} = 0.65 \cdot \left(\frac{t_p}{d}\right)^{1.35} \cdot \left(\frac{p}{d}\right)^{-0.26} \qquad \left(\frac{t_p}{d} > 0.8\right)$$
(3-25)

where

d = channel diameter [m]

 t_p = plate thickness [m]

p = horizontal pitch [m]



Figure 3-2. PCHE channel configurations.

3.4 Leakage

The leak rate of helium from the loop, $R_{leak,He}$ [m³ (STP)/s], can be expressed as

$$R_{leak,He} = V_{total} \cdot L_R \tag{3-26}$$

where

 V_{total} = total inventory in loop [m³ (STP)]

 L_R = helium leak rate [1/s].

The leak rate of H₂, HT, and HTO with helium leakage in node j, $R_{leak,i,j}$, is calculated as

$$R_{leak,i,j} = V_{total} \cdot L_R \cdot \overline{C}_i \cdot \frac{V_j}{V_{leak}} \quad (i = H_2, \text{ HT and HTO})$$
(3-27)

where

 \overline{C}_i = average concentration of chemical i in nodes with helium leak

 V_{leak} = sum of the inventory of nodes with helium leak.

3.5 Purification

The removal rate, R_{PF} , of hydrogen and tritium-containing chemicals in the primary and intermediate loop by the purification system is expressed as

$$R_{PF,i,j} = F_{PF,He} \cdot \eta_i \cdot C_{i,j} \quad (i = H_2, HT, and HTO)$$
(3-28)

where

 $F_{PF,He}$ = helium flow rate at purification system [m³ (STP)/s]

 η_i = fractional efficiency of purification system for removing component i.
3.6 Electrolyzer

A very simple electrolyzer model has been implemented in TPAC in order to match the species conservation in the following chemical reaction:

$$H_2O = H_2 + 0.5O_2 \tag{3-29}$$

In this model, a complicated electrochemistry model has not been applied. Decomposition of H_2O into H_2 and O_2 were simply considered by conversion ratios. HTO was considered to be chemically identical to H_2O , so the same conversion correlation was applied to HTO decomposition. Produced O_2 in the cathode was considered to be directly transferred to the anode.

3.7 Isotope Exchange

The SI process contains H_2 , H_2O , H_2SO_4 , and HI and the HTE process contains H_2 and H_2O as hydrogen-containing chemicals. The primary and intermediate coolants also contain H_2 and H_2O as impurities. To analyze the tritium behaviors in the VHTR/hydrogen production system, we need some additional models that can handle isotope exchange reactions between some chemicals. The TPAC takes into account the following isotope exchange reactions between tritium containing chemicals and hydrogen containing chemicals:

- Reaction-1: $HT + H_2O = H_2 + HTO$
- Reaction-2: $HT + H_2SO_4 = H_2 + HTSO_4$
- Reaction-3: $HT + HI = H_2 + TI$
- Reaction-4: $HTO + H_2SO_4 = H_2O + HTSO_4$
- Reaction-5: $HTO + HI = H_2O + TI$.

Chemical equilibrium of the isotope exchange reaction from Reaction-1 through Reaction-5 can be expressed by the following equations.

$$K_{1} = \frac{F_{H2,eq} \cdot F_{HT0,eq}}{F_{HT,eq} \cdot F_{H20,eq}}$$
(3-30)

$$K_2 = \frac{F_{H2,eq} \cdot F_{HTSO4,eq}}{F_{HT,eq} \cdot F_{H2SO4,eq}}$$
(3-31)

$$K_3 = \frac{F_{H2,eq} \cdot F_{TI,eq}}{F_{HT,eq} \cdot F_{HI,eq}}$$
(3-32)

$$K_{4} = \frac{K_{2}}{K_{1}} = \frac{F_{H2O,eq} \cdot F_{HTSO4,eq}}{F_{HTO,eq} \cdot F_{H2SO4,eq}}$$
(3-33)

$$K_{5} = \frac{K_{3}}{K_{1}} = \frac{F_{H2O,eq} \cdot F_{TI,eq}}{F_{HTO,eq} \cdot F_{HI,eq}}$$
(3-34)

where

K = equilibrium constant of each isotope exchange reaction

 $F_{i,eq}$ = volumetric flow rate of each chemical at equilibrium [m³(STP)/s]

In the above equations, Reaction-4 and Reaction-5 can be expressed by combinations of Reaction-1 through Reaction-3. Therefore, equations (3-33) and (3-34) do not need to be solved. To solve Equations (3-30), (3-31), and (3-32), the equilibrium (the values of $F_{i,eq}s$ were expressed as:

$$F_{HT,eq} = F_{HT,IN} - \Delta_1 - \Delta_2 - \Delta_3 \tag{3-35}$$

$$F_{H2,eq} = F_{H2,IN} + \Delta_1 + \Delta_2 + \Delta_3$$
(3-36)

$$F_{H20,eq} = F_{H20,IN} - \Delta_1$$
(3-37)

$$F_{HTO,eq} = F_{HTO,IN} + \Delta_1 \tag{3-38}$$

$$F_{H2SO4,eq} = F_{H2SO4,IN} - \Delta_2 \tag{3-39}$$

$$F_{HTSO4,eq} = F_{HTSO4,IN} + \Delta_2 \tag{3-40}$$

$$F_{HI,eq} = F_{HI,IN} - \Delta_3 \tag{3-41}$$

$$F_{TI,eq} = F_{TI,IN} + \Delta_3 \tag{3-42}$$

where

$$F_{i,IN}$$
 = inlet volumetric flow rate of each chemical species [m³(STP)/s]

 Δ_n = amount of HT conversion rate by Reaction-n [m³(STP)/s].

From the above set-up, we can solve three unknowns $(\Delta_1, \Delta_2, \Delta_3)$ from three equations (Equations (3-30), (3-31) and (3-32)).

The equilibrium constants of Reaction-1 through Reaction-3 can be expressed in the following equation (Atomic Energy Society of Japan 1982):

$$\log K_1 = 0.292 \log T + (336.5/T) - 1.055 \tag{3-43}$$

 $\log K_2$: unknown

(3-44)

$$\ln K_3 = -1.13 + 1.07 \times 10^{-2} T - 2.26 \times 10^{-5} T^2 + 2.10 \times 10^{-8} T^3 - 7.25 \times 10^{-12} T^4$$
(3-45)

4. TRITIUM PERMEATION ANALYSES CODE (TPAC)

4.1 Program Overview

The Tritium Permeation Analyses Code (TPAC) is the computer code developed by Idaho National Laboratory (INL) for the purpose of analyses of tritium behaviors and distributions in the VHTR systems including integrated hydrogen production systems such as HTE and SI processes. The TPAC is based on the mass conservation equations of various tritium-containing species (HT, HTO, HTSO₄, and TI) and hydrogen (H₂) coupled with a variety of tritium source and sink models. The models include tritium sources from ternary fission and neutron reactions, leakage, permeation through solid walls, purification, isotope exchange reactions, and electrolysis. The detailed models are well described in Section 3. The TPAC provides large flexibility for system configurations and very convenient graphical user interface (GUI). The system can be configured (not only for VHTR but also general components) using easy dragand-drops on the screen.

The TPAC has been developed using MATLAB SIMULINK Packages (MathWorks, 2009), a world famous numerical computing environment and language. Currently, more than one million people are reportedly using this software across many industries and universities. MATLAB is a high-level technical computing program language created by MathWorks. It provides easy matrix manipulation, algorithms implementation, user interface creation, and numerous built-in numerical libraries. SIMULINK is a built-in package integrated into MATLAB that specializes in modeling, simulating, and analyzing multidomain dynamic systems. Its primary interface is a graphical block diagramming tool and a customizable set of block libraries. It offers tight integration with the rest of the MATLAB environment. It supports linear and nonlinear system modeling in continuous time, sampled time, or a hybrid of both.

The TPAC has been developed based on the SIMULINK user interface. All TPAC component libraries were built in the SIMULINK model library so the TPAC component blocks can be easily integrated with preexisting SIMULINK model blocks. It provides enormous capabilities for code flexibility and extensibility. Figure 4-1 shows the basic TPAC user interface integrated in MATLAB SIMULINK. Therefore, the TPAC interface is exactly the same as MATLAB SIMULINK interface except for the TPAC process component library, which was developed and designed for tritium analyses in this work. Symbol ^① in Figure 4-1, is the default MATLAB Workspace where the entire status of the simulation can be checked. The data calculated in the SIMULINK Workspace can be interacted and postprocessed. Symbol ^② is the SIMULINK Workspace where all the component blocks were arranged from the TPAC libraries for submitting the job with some detailed user settings. Symbol ^③ is the SIMULINK Library Browser in which the TPAC libraries are implemented and integrated. Users can easily find the TPAC components in this browser and move them to the SIMULINK Workspace for making block models. The following sections will describe the TPAC structures and how to develop and run the model.

4.2 TPAC Structures

This section describes the TPAC code structures and how to install and start the program. Figure 4-2 shows the TPAC organization. In the TPAC folder, there are two files and two folders. The file, 'TPAC_BETA_Lib.mld.' is the main program containing all the component blocks. The file, "slblocks.m" is the file for linking the TPAC program and the MATLAB SIMULINK; without this file, the TPAC component blocks do not show in the SIMULINK Library Browser. The folder 'functions' includes some function libraries related to tritium-related models such as isotope exchange reaction. The folder 'Icons' includes all the icons used in the TPAC GUI.



Figure 4-1. Basic TPAC user interface integrated into MATLAB SIMULINK.



Figure 4-2. TPAC organizations.

Installation of the TPAC is very user-friendly, but because the code is developed based on sharing the SIMULINK interface, MATLAB SIMULINK Package should be installed first. The basic MATLAB SIMULINK package (student edition) is sufficient to install and run the TPAC. The following instruction directs how to install the TPAC in the MATLAB SIMULINK environment.

TPAC Installation

- 3. Copy 'TPAC' Folder in any location where the user wants to place it.
- 4. Run MATLAB

Just double-clicking the MATLAB icons can make the MATLAB program start. Figure 4-3 shows the default Workspace of MATLAB.

MATLAB	
File Edit Debug Desktop Window H	telp
D 🗃 👗 🖿 🛍 က က 🕷 🖆	Current Directory: C:'MATLAB701 work
Shortcuts 🗷 How to Add 💽 What's New	
Workspace 7 ×	Command Window 🛷 🛪
📸 📑 趣 🔜 🔻 Base 👻 👋	
Name ∠ Value	To get started, select MATLAB Help or Demos from the Help menu.
Image: Command History I	<pre>Warning: Unable to register DirectX component C:\MATLAB701\toolbox\matl Warning: Unable to register DirectX component C:\MATLAB701\toolbox\matlab\audiov > In initprefs at 34 In matlabrc at 108 >> </pre>
A Start	

Figure 4-3. MATLAB Workspace.

5. Set Path of the TPAC Folder

To use the TPAC in the MATLAB, users should set the path of the TPAC so that the MATLAB program recognizes that the TPAC is integrated into the MATLAB environment. To add the path of the TPAC, click File/Set path in the main menu (See Figure 4-4) to open the set path window (See Figure 4-5). The TPAC folder can be added by clicking the 'Add with subfolders ...' button and including the TPAC folder that includes all the related files as shown in Figure 4-5. Adding the TPAC folder in the path automatically integrates the TPAC into the MATLAB Library.



Figure 4-4. Set path of the MATLAB.

📣 Set Path		
All changes take effect imme	ediately.	
	MATLAB search path:	
Add Folder	C:\MATLAB701\work\tpac betatest	<u> </u>
Add with Subfolders	C:\MATLAB701\work\tpac\fucntions	
click! Move to Top	C:\MATLAB701\work\tpac\icons	
Move Up	C:\MATLAB701\toolbox\matlab\general	
Move Down	C:\MATLAB701\toolbox\matlab\lang	
Move to Bottom	C:\MATLAB701\toolbox\matlab\elfun C:\MATLAB701\toolbox\matlab\specfun	
Remove	C:\MATLAB701\toolbox\matlab\matfun	*
Save Close	Revert Default	Help

Figure 4-5. Set path windows in MATAB.

6. Check the Linkage. The linkage can be easily checked if the TPAC is successfully integrated into the MATLAB by running SIMULINK. The SIMULINK can be started by clicking the following button in the MATLAB toolbar.

👣 : Starting SIMULINK in MATLAB

When the above button is clicked, the SIMULINK Library Browser will pop up as shown in Figure 4-6. The TPAC can be seen in the Model List on the left-hand side, and the TPAC components can be seen in the Component List on the right-hand side. As shown in this figure, the current version is TPAC 1.5b.



Figure 4-6. TPAC Library Browser.

TPAC Structures

As shown in Figure 4-7, the TPAC has a variety of library categories. Each category has different sets of component block models. Figure 4-7 shows the structure of the TPAC component libraries. TPAC 1.5b consists of 12 different categories. **VHTR System Component** contains the component blocks related to the VHTR systems such as core and vessel. **Pipe Component** contains various types of pipe component blocks including VHTR, HTSE, and SI systems. **Heat Exchanger Component** contains two different heat exchanger component blocks. **Vessel Component** contains various vessel components related to VHTR, HTSE, and SI systems. **Containment Component** contains the reactor containment block. **Purification System Component** contains a tritium purification system block. **Electrolyzer Component** contains an electrolyzer block. **Flow Distributions Component** contains flow distributor and mixer blocks. **Detector Component** contains a detector block that can be used for open-flow or fixed flows. **Detector Component** contains data save blocks that export calculated results to MATLAB Workspace or external files. **Sample Simulation** contains some sample simulation cases, including those used for code verification works. Details on the each component in the TPAC are will described in Section 4.4.



Figure 4-7. TPAC library categories.

4.3 Starting TPAC for Tritium Analyses

This section shows a simple application of the TPAC into VHTR tritium analyses to understand how to develop a TPAC model and analyze it. This modeling application includes (1) starting TPAC, (2) manipulating TPAC components, (3) linking components, (4) implementing input data, (5) setting solver and running, and (6) post-processing results. Figure 4-8 shows the sample system configuration for tritium analyses. This system consists of primary and secondary coolant loops. The primary loop has a reactor, pipes (hot leg and cold leg), a compressor, and a purifier. The secondary loop has a power conversion unit and a secondary purifier. The primary and the secondary loops are connected by an IHX. All the systems are enclosed in the containment building. Tritium in the primary side was assumed to have penetrated through the IHX. Leakage of coolant from the primary side to the containment has been assumed through the cold leg and the reactor vessel. No leakage has been considered in the secondary loops. This system does not represent any specific reactor designs and therefore the parameters and the input data were arbitrarily chosen. Therefore, the results obtained in this sample simulation have no practical meaning for the actual reactor analyses. Figure 4-9 shows the final TPAC modeling for the sample configuration developed in this section.



Figure 4-8. Sample system configuration for tritium analyses.



Figure 4.9. TPAC sample system simulation.

4.3.1 Starting TPAC

Starting the TPAC is also very user-friendly. The following is the procedure to start a new project of the TPAC.

- 1. Run MATLAB. Double-clicking **MATLAB icon** on the window desktop will start the MATLAB program and show up the MATLAB workspace.
- 2. Run SIMULINK. Clicking **SIMULINK button** on the MATLAB toolbar will show up the SIMULINK Library Browser (See Figure 4-10).
- 3. Start a New TPAC Project (SIMULINK Editor). By clicking the **New button** on the Library Browser, a new TPAC project (SIMULINK Editor) can be started (See Figure 4-10).
- 4. Save a File Name. The new project can be saved from 'File/Save' on the Menu.



Figure 4-10. Start a new TPAC project.

4.3.2 Developing Input Diagram

To simulate the tritium behaviors in the VHTR systems, the input file that consists of block diagrams should be developed first. The TPAC is based on the GUI which provides easy drag-and-drop features for developing a model diagram. This section describes how to manipulate the model components and how to develop the input model diagram with those components.

Basic Manipulation of the Component Blocks

Select an object by clicking it. Small black square "handles" appear at the corners of a selected block and near the end points of a selected line. For example, Figure 4-11 shows a selected reactor component, containment component, and pipe component. When users select an object by clicking it, any other selected objects are deselected.



Figure 4-11. Selected TPAC model components.

More than one object can be selected either by selecting objects one at a time, by selecting objects located near each other using a bounding box, or by selecting the entire model. To select more than one object by selecting each object individually, hold down the **Shift key** and click each object to be selected. To deselect a selected object, click the object again while holding down the Shift key. To select all objects in the active window, choose **Select All** from the **Edit** menu.

TPAC component or SIMULINK block diagrams use lines to represent pathways for signals among blocks in a model. TPAC (or SIMULINK) can connect blocks by drawing lines from their output ports to their input ports. Figure 4-12 shows a simple example of connecting Reactor and Pipe components. First, click the output port of the reactor, and then drag the line to the input port of the pipe. Figure 4-12 indicates that the reactor outflow is coming into the pipe inflow.



Figure 4-12. Connection of components (between Reactor and Pipe).

Input Diagram Development for Sample Configuration

The input diagram shown in Figure 4-9 will be developed through this section. The following is the procedure for developing the input diagram.

- Place Reactor Component in the SIMULINK Editor. The Reactor component block (TPAC 1.5b/A. VHTR System Component/Reactor) can be moved from the library browser to the editor by drag-anddrop. First, click and hold Reactor block in the library browser (mouse left button), and then drag mouse to the editor. When the mouse pointer is located where the users wants, drop the component by releasing mouse button. Figure 4-13 shows the procedure.
- 2. Place all other components into the editor. As shown above, the other components can also be placed in the editor by the same methods. Place and arrange all the components as shown in Figure 4-14. Table 4-1 summarizes the components and their locations. The name of the component can be easily changed in the editor.
- 3. Link all the components. After placing the components, users should link all the components to interchange their information and signal; the linking method was already explained in Figure 4-12. By using the linking method, the model components can be linked as shown in Figure 4-15).
- 4. Add detector components. To read or plot the data after simulation, detector components should be connected to the streams. The detector component can be found in TPAC 1.5b/J. Detector Component/ Detector of the library browser. Figure 4-16 shows how to add a detector component into the editor. Like the other components, the detector block can be added using the drag-and-drop feature. After placing the block in the editor, the size of the detector can be adjusted easily. To link the detector and the stream, users need a different method as shown in Figure 4-17. First, click the stream of interest with ctrl button, another branch line is then created that is connected to the main stream. This line can be connected to the detector block. Using the same method, detector blocks can be added to all the streams as shown in Figure 4-18. The detector blocks allow the user to access the data in the connected streams during or after simulation.



Figure 4-13. Moving reactor component from library browser to editor.



Figure 4-14. Component arrangement for the sample configuration.

Component Name	Locations in Library Browser
Reactor	TPAC 1.5b/A. VHTR System Component/ Reactor
Tritium Source	SIMULINK/Source/ Constant
Hot Leg	TPAC 1.5b/B. Pipe Component/ Pipe-VHTR
Cold Leg	TPAC 1.5b/B. Pipe Component/ PipeII-VHTR
Heat Exchanger	TPAC 1.5b/C. Heat Exchanger Component/Heat Exchanger
PCU	TPAC 1.5b/B. Pipe Component/ Pipe-VHTR
Containment	TPAC 1.5b/E. Containment Component/ Containment
Purification System 1	TPAC 1.5b/F. Purification System Component/ Purification System
Purification System 1	TPAC 1.5b/F. Purification System Component/ Purification System





Figure 4-15. Linked model components.



Figure 4-16. Detector addition into the editor.



Figure 4-17. Linking the detector block into the stream.



Figure 4-18. Addition of detectors to the main streams.

4.3.3 Setting Up Input Parameters

Once users develop the input diagram of the simulation, the input parameters should be defined for each component block. The current section summarizes the parameters defined in this simulation. The input data used in this simulation were arbitrarily chosen, so it does not represent any specific reactor designs such as gas turbine modular helium reactor or High Temperature Test Reactor. To access the input parameters, users can **double-click** the component blocks. Each block has its own input windows that represent its component types. The following summarizes the input data defined in the components.

Reactor

Figure 4-19 shows the input windows for the **Reactor** component. Specify each parameter as shown in this figure.

Function Block Parameters: Reactor	
VHTR-Core (mask)	٦
This block is desined for solving tritium behaviors in the VHTR reactor core.	
Developer: Eung Soo Kim (INL)	
Parameters	
Channel Volume [m3] (Channel 1)	
30	
Temperature [K] (Channel 1)	
800	
Pressure [Pa] (Channel 1)	
500000	
Flow Rate [m3(STP)/s]	
1	
HT Initial Concentration (Channel 1)	
1e-50	
H2 Initial Concentration (Channel 1)	
1e-5	
└ Leakage (Yes/No)	
Permeation to Outside (Yse/No)	
<u>O</u> K <u>C</u> ancel <u>H</u> elp <u>Apply</u>	

Figure 4-19. Input windows for Reactor component.

Tritium Source

Figure 4-20 shows the input windows for the **tritium source** block. Specify each parameter as shown in this figure.

Constant Output the constant specified by the 'Constant value' parameter. If 'Constant value' is a vector and 'Interpret vector parameters as 1-D' is on, treat the constant value as a 1-D array. Otherwise, output a matrix with the same dimensions as the constant value. Main Signal data types Constant value: 1.0e-12 Image: Interpret vector parameters as 1-D Sample time: inf Inf OK QK Cancel	Source Block Parameters: Tritium Source (m3(STP)/s)
Output the constant specified by the 'Constant value' parameter. If 'Constant value' is a vector and 'Interpret vector parameters as 1-D' is on, treat the constant value as a 1-D array. Otherwise, output a matrix with the same dimensions as the constant value. Main Signal data types Constant value: [1.0e-12] ✓ Interpret vector parameters as 1-D Sample time: [inf OK Cancel	Constant
Main Signal data types Constant value:	Output the constant specified by the 'Constant value' parameter. If 'Constant value' is a vector and 'Interpret vector parameters as 1-D' is on, treat the constant value as a 1-D array. Otherwise, output a matrix with the same dimensions as the constant value.
Constant value: 1.0e-12 ✓ Interpret vector parameters as 1-D Sample time: Inf <u>OK</u> <u>Cancel</u> <u>H</u> elp	Main Signal data types
1.0e-12 ✓ Interpret vector parameters as 1-D Sample time: inf QK Cancel	Constant value:
✓ Interpret vector parameters as 1-D Sample time: inf QK Cancel	1.0e-12
Sample time: inf OK Cancel Help	Interpret vector parameters as 1-D
inf <u> QK</u> <u>Cancel</u> <u>H</u> elp	Sample time:
OK Cancel Help	inf
<u>O</u> K <u>C</u> ancel <u>H</u> elp	
	<u>Q</u> K <u>C</u> ancel <u>H</u> elp

Figure 4-20. Input windows for Tritium Source block.

Hot Leg

Figure 4-21 shows the input windows for the **Hot Leg** component. Specify each parameter as shown in this figure.

Function Block Parameters: Hot Leg
VHTR-Pipe (mask)
This component is designed for analysis of tritium behaviors in the VHTR pipe system.
Parameters
Channel Volume [m3] (Channel 1)
10
Temperature [K] (Channel 1)
800
Pressure [Pa] (Channel 1)
200000 Free Brite (=2/07TR) (-1
How Rate [m3(51P)/s]
HT Initial Concentration (Channel 1)
1e-50
H2 Initial Concentration (Channel 1)
1e-5
🗌 Leakage (Yes/No)
Permeation to Outside (Yse/No)
<u>QK</u> ancel <u>H</u> elp <u>Apply</u>

Figure 4-21. Input windows for Hot Leg component block.

Cold Leg

Figure 4-22 shows the input windows for the **Cold Leg** component. Specify each parameter as shown in this figure.

Function Block Parameters: Cold Leg
VHTR-Pipe (II) (mask)
This component is designed for analysis of tritium behaviors in the VHTR pipe system. External Leak and Permeation In/Out can be considered.
Parameters
Channel Volume [m3] (Channel 1)
10
Temperature [K] (Channel 1)
500
Pressure [Pa] (Channel 1)
500000
Flow Rate [m3(STP)/s]
1
HT Initial Concentration (Channel 1)
1e-50
H2 Initial Concentration (Channel 1)
1e-5
✓ Leakage (Yes/No)
Total Inventory in Loop [m3(STP)]
1000
Fractional Helium Leak Rate [1/s]
1.15e-7
Sum of the Inventory of Nodes with Helium Leak [m3(STP)]
200
Permeation to Outside (Yse/No)
OK Cancel Help Apply

Figure 4-22. Input windows for Cold Leg component block.

PCU

Figure 4-23 shows the input windows for the **PCU** component. Specify each parameter as shown in this figure.

Function Block Parameters: PCU	
VHTR-Pipe (mask)	7
This component is designed for analysis of tritium behaviors in the VHTR pipe system.	
Parameters	
Channel Volume [m3] (Channel 1)	
50	
Temperature [K] (Channel 1)	
800	
Pressure [Pa] (Channel 1)	
801300	
Flow Rate [m3(STP)/s]	
1	
HT Initial Concentration (Channel 1)	
0	
H2 Initial Concentration (Channel 1)	
1e-20	
Leakage (Yes/No)	
Permeation to Outside (Yse/No)	
OK <u>Cancel</u> <u>Help</u> <u>Apply</u>	

Figure 4-23. Input windows for PCU component block.

Purification System 1

Figure 4-24 shows the input windows for the **Purification System 1** component. Specify each parameter as shown in this figure.

Function Block Parameters: Purification System1
Purification System (mask)
This component is designed for calculating purification system.
Parameters
Fraction of Mass Flow to Purification System (0~1)
5e-2
Fractional Efficiency for HT Purification (0~1)
1
Fractional Efficiency for H2 Purification (0~1)
10
Fractional Efficiency for H2O Purification (0~1)
Fractional Efficiency for HTO Purification (0~1)
lo.
<u>O</u> K <u>Cancel</u> <u>H</u> elp <u>Apply</u>

Figure 4-24. Input windows for Purification System 1 component block.

Purification System 2

Figure 4-25 shows the input windows for the **Purification System 2** component. Specify each parameter as shown in this figure.

Function Block Parameters: Purification System2
Purification System (mask)
This component is designed for calculating purification system.
Parameters
Fraction of Mass Flow to Purification System (0~1)
5e-2
Fractional Efficiency for HT Purification (0~1)
1
Fractional Efficiency for H2 Purification (0~1)
0
Fractional Efficiency for H2O Purification (0~1)
0
Fractional Efficiency for HTO Purification (0~1)
0
<u>QK</u> <u>Cancel</u> <u>H</u> elp <u>A</u> pply

Figure 4-25. Input windows for Purification System 2 component block.

Containment

Figure 4-26 shows the input windows for the **Containment** component. Specify each parameter as shown in this figure.

Function Block Parameters: Containment
VHTR-Containment (mask)
This component is designed for analysis for tritium behaviors in the VHTR containment.
Parameters
Channel Volume [m3] (Channel 1)
10000
Temperature [K] (Channel 1)
323
Pressure [Pa] (Channel 1)
1.57e5
Flow Rate [m3(STP)/s]
ļO
HT Initial Concentration (Channel 1)
JO
H2 Initial Concentration (Channel 1)
Eleakage (Tes/No)
Ferneduor to Outside (158/190)
QK <u>Cancel</u> <u>H</u> elp <u>Apply</u>

Figure 4-26. Input windows for Containment component block.

Heat Exchanger

Figure 4-27 shows the input windows for the **Heat Exchanger** component. Specify each parameter as shown in this figure.

SE-Heat Exchanger (mask)	
his component is designed for analysis of tritium behaviors in the HTSE heat cchanger systems.	
rameters	
Ion Exchange Model (Yes/No)	
hannel Volume [m3] (Channel 1)	
10	
emperature [K] (Channel 1)	
50	
ressure [Pa] (Channel 1)	
500000	
(T Initial Concentration (Channel 1)	
le-50	
2 Initial Concentration (Channel 1)	
le-5	
hannel Volume [m3] (Channel 2)	
10	
emperature [K] (Channel 2)	
350	
ressure [Pa] (Channel 2)	
500000	
12 Initial Concentration (Channel 2)	
1e-20	
leat Transfer Area [m2]	
1	
tx Wall Thickness [m]	
0.01	
IX Wall Temperature [K]	
273	1
re-exponential Factor for HT Permeability	
1.36e-10	
re-exponential Factor for H2 Permeability	
0	
ctivation Energy for HT Permeation [J/mol K]	
53.5e3	- î
ctivation Energy for H2 Permeation [J/mol K]	
74e3	

Figure 4-27. Input windows for Heat Exchanger component block.

4.3.4 Running Job

In the previous section, the input diagram has been developed and the input parameters have been specified. So the model is now ready for running. Before running the simulation, several parameters related to the solver should be predefined. The parameters include solver type, maximum/minimum time step, and initial time step. In this simulation, the default setting has been used so that any changes of the

setting are not required. The details for the solver setting are described Section 4.5. Figure 4-28 shows the input diagram of the sample simulation. Running the TPAC simulation involves the following steps:

- 1. Enter the calculation time into the box in the toolbar, which is specified in Figure 4-28 as 1e4 sec.
- 2. Click the **Run** button on the toolbar; the simulation will now perform from 0 to 1e4 sec.
- 3. Click the **Stop** button to terminate the simulation.



Figure 4-28. Running TPAC simulation.

4.3.5 Analyses on the Results

After simulation, the results can be easily checked by detector blocks that have been already linked to the streams. The detector block shows the upstream flow conditions. For example, the **D1** detector block contains the information of the Hot Leg component. The data of the detector can be accessed by **double-clicking** the icons. Figure 4-29 shows the **D5** detector data which contains the Reactor component information. The detector block consists of two parts; display and plot. **Display** region shows the data of the stream at the end of calculations. **Plot** region displays the graph of time vs. specific data. The data are arranged in the following sequence from top to bottom.

- Flow Rate [m³(STP)/s]
- HT Concentration [mole fraction]

- Temperature [K]
- Pressure [Pa]
- H₂ Concentration [mole fraction]
- H₂O Concentration [mole fraction]
- HTO Concentration [mole fraction]
- H₂SO₄ Concentration [mole fraction]
- HTSO₄ Concentration [mole fraction]
- HI Concentration [mole fraction]
- TI Concentration [mole fraction]



Figure 4-29. Inside of Detector block (D5).

According to the simulation results, the HT concentration in the Reactor is 1.976×10^{-11} mole fraction (=0.01976 ppb) after 1×10^4 sec (=2.8 hrs). Figure 4-30 shows the variations of the HT concentrations. This plot can be popped up by double-clicking the **Plot** icons. Initially, the HT concentration rapidly increases, it then becomes saturated by the balance of generation (tritium source) and removal (purification system, leakage, and permeation).



Figure 4-30. Variations of the HT concentrations in the reactor.

Figures 4-31 and 4-32 show the HT concentrations in the PCU (**D8** detector) and the containment (**D9** detector), respectively. The HT concentrations in the PCU are initially increased by permeation of the tritium through the heat exchanger (See Figure 4-31). They are then getting saturated by purification of the HT. In the containment, the HT concentrations are continuously increased by leakage of coolant from the primary loop (See Figure 4-32). No leakage, permeation, or purification was considered in the containment and therefore HT concentrations are continuously increased. For the other components or streams, users can also access the data as described above.



Figure 4-31. Variations of the HT concentrations in the PCU.



Figure 4-32. Variations of the HT concentrations in the containment.

4.4 Component Summary

The TPAC consists of a variety of model blocks that represent VHTR or hydrogen production system components. This section summarizes the type of component blocks and how to use them. Table 4-2 lists the component organizations in the TPAC. The TPAC has 11 categories, each consisting of its own component blocks that represent reactor components, hydrogen production system components, or other subsystems. The details about these components are described in the following subsections.

4.4.1 VHTR System Component

VHTR System Component category has two component blocks; **Reactor** and **Core**. These two components are used to model the VHTR core and the pressure vessel.

Reactor Component

Table 4-3 summarizes the specifications of the **Reactor** block. This table shows the governing equations and models used in this block. It also shows port information needed for actual modeling in the SIMULINK editor. The **Reactor** block solves two species conservation equations (HT, H₂) with leakage and permeation models. This block has four ports that exchange information with other component blocks. Table 4-4 lists the input parameters that should be defined in the reactor model.

Core

Table 4-5 shows the specifications of the **Core** block. This block calculates various tritium generations from the core such as ternary fission and neutron reactions. Figure 4-33 shows the inside of the core block. It basically consists of five different tritium source calculation blocks. However, depending on the user's choice, it can be very flexibly modified to represent various designs by simply adding or deleting components.

Category	С	Component Blocks	
		Ternary Fission	
		Birth from 6Li	
		Birth from 7Li	
	Core	Birth from 3 He	
A. VHTR System Component		Birth from 10B	
		Convert dNT/dt to Rcore	
		Mux	
	Reactor		
	Pipe-VHTR		
	Pipe-HTSE		
	Pipe-SI		
	Pipe II-VHTR		
B. Pipe Component	Pipe II-HTSE		
	Pipe II-SI		
	Isotope Exchange		
	External Tritium Source		
	Terminator		
	Heat Exchanger		
C. Heat Exchanger Component	Heat Exchanger	II	
	Vessel-VHTR		
	Vessel-HTSE		
	Vessel-SI		
D. Vessel Component	Isotope Exchang	e	
	External Tritium Source		
	Terminator		
E. Containment Component	Containment		
F. Purification System Component	Purification System		
G. Electrolyzer Component	Electrolyzer		
II Flow Distribution Component	Flow Splitter		
H. Flow Distribution Component	Flow Distributor		
I. Boundary Flow Component	Boundary Flow		
J. Detector Component	Detector		
K. Doto Sava Component	Save to File		
K. Data Save Component	Save to Workspace		

Table 4-2. Component organization of the TPAC.

Icon		Specifications
		HT Species Conservation
49298-	Governing	H ₂ Species Conservation
3	Equations and Models	Leakage Model
	Widdels	Permeations Model
1 Pc Reactor	Port Information	1 Flow In
		2 Flow Out
		3 Tritium Source (Core Model) In
		Iritium Leak and Permeation Out

Table 4-3. Specifications of the Reactor Component block.

Table 4-4. Input parameters of the Reactor Component block.

Input Parameter	Unit	Description
Channel volume	m ³	Volume of the reactor
Temperature	Κ	Average temperature in the reactor
Pressure	Ра	Average pressure in the reactor
Flow rate	m ³ (STP)/s	Flow rate of the reactor outlet
HT initial concentration	mole fraction	Average initial HT mole fraction in the reactor
H ₂ initial concentration	mole fraction	Average initial H ₂ mole fraction in the reactor
Leakage	#	Check box for activating leakage model
Total inventory in loop	m ³ (STP)	Total inventory of the loop including the reactor
Fractional helium leak rate	1/s	Fraction helium leak rate
Sum of the inventory of the nodes with helium leak	m ³ (STP)	Sum of the inventory of the loop with helium leak
Permeation to outside	#	Check box for activating permeation model
Surface area for permeation	m ²	Surface area for permeation in the reactor
Thickness of the wall	m	Wall thickness for the permeation in the reactor
Permeation temperature	Κ	Temperature for permeation
Pre-exponential factor for HT permeation	m ³ (STP)/m/s	Pre-exponential factor for HT permeation
Activation energy for HT permeation	J/mol	Activation energy for HT permeation
Pre-exponential factor for H ₂ permeation	m ³ (STP)/m/s	Pre-exponential factor for H ₂ permeation
Activation energy for H ₂ permeation	J/mol	Activation energy for H ₂ permeation



Table 4-5	Specification	of Core Com	nonent block
1 able + 3.	Specification	of core com	ponent block.

Figure 4-33. Core block structure.

4.4.2 Pipe Component

The **Pipe Component** category has nine component blocks as shown in Table 4-2. Tables 4-6 through 4-18 summarize specifications and input parameters of the components related to this category.

Tables 4-6 and 4-7 show the specifications of two VHTR-Pipe components. The two components are composed of the same equation sets, but they have different user interfaces. **Pipe-VHTR** has only **Flow In** and **Flow Out** ports in the block while **PipeII-VHTR** has two additional ports: **Tritium Leak and Permeation In/Out ports**. The **Tritium Leak and Permeation Out** port of a pipe component can be connected to the **Tritium Leak and Permeation** in the port of another pipe, vessel, or containment block. Tables 4-9, 4-10, and 4-11 summarize the **HTSE-Pipe** related component blocks. These blocks solve two additional species conservation equations (H₂O, HTO) than the **VHTR-Pipe** blocks. Tables 4-12, 4-13, and 4-14 summarize the **SI-Pipe** blocks. These blocks solve six more species conservation equations (H₂O, HTO, H₂SO₄, HTSO₄, HI, and TI) compared to the **VHTR-Pipe** blocks. Tables 4-16 and 4-17 summarize the **Isotope Exchange** block, which provides three isotope exchange reactions. Tables 4-18 and 4-19 summarize the **External Tritium Source** block, which provides external user input for tritium source. This block can be easily connected to the **Tritium Leak and Permeation In** port in the pipe blocks.

Icon		Specifications
		HT Species Conservation
	Governing	H ₂ Species Conservation
1 > 2 Pipe - VHTR	Equations and Models	Leakage Model
		Permeations Model
	Port Information	1 Flow In
		2 Flow Out

Table 4-6. Specification of Pipe-VHTR component block.

Table 4-7. Specification of PipeII-VHTR component block.

Icon		Specifications	
	Governing Equations and Models	HT Species Conservation	
		H ₂ Species Conservation	
		Leakage Model	
3 1 Pipe II - VHTR 2		Permeations Model	
	Port Information	1 Flow In	
		2 Flow Out	
		3 Tritium Leak and Permeation In	
		4 Tritium Leak and Permeation Out	

Input Parameter	Unit	Description
Channel Volume	m ³	Volume of the pipe
Temperature	Κ	Average temperature in the pipe
Pressure	Ра	Average pressure in the pipe
Flow Rate	m ³ (STP)/s	Flow rate of the pipe outlet
HT Initial Concentration	mole fraction	Average initial HT mole fraction in the pipe
H ₂ Initial Concentration	mole fraction	Average initial H ₂ mole fraction in the pipe
Leakage	#	Check box for activating leakage model
Total Inventory in Loop	m ³ (STP)	Total inventory of the loop including the pipe
Fractional Helium Leak Rate	1/s	Fraction helium leak rate
Sum of the inventory of the nodes with helium leak	m ³ (STP)	Sum of the inventory of the loop with helium leak
Permeation to Outside	#	Check box for activating permeation model
Surface Area for Permeation	m ²	Surface area for permeation in the pipe
Thickness of the Wall	m	Wall thickness for the permeation in the pipe
Permeation Temperature	K	Temperature for permeation
Pre-exponential Factor for HT Permeation	m ³ (STP)/m/s	Pre-exponential factor for HT permeation
Activation Energy for HT Permeation	J/mol	Activation energy for HT permeation
Pre-exponential Factor for H ₂ Permeation	m ³ (STP)/m/s	Pre-exponential factor for H ₂ permeation
Activation Energy for H ₂ Permeation	J/mol	Activation energy for H ₂ permeation

Table 4-8. Inp	ut Parameters of Pi	pe-VHTR com	ponent block.
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Table 4-9. Specification of Pipe-HTSE component block.

Icon		Specifications
		HT Species Conservation
		H ₂ Species Conservation
	Governing	H ₂ O Species Conservation
1 > 2 Pipe - HTSE	Equations and Models	HTO Species Conservation
		Leakage Model
		Permeations Model
	Port Information	1 Flow In
		2 Flow Out

Icon		Specifications
		HT Species Conservation
		H ₂ Species Conservation
	Governing	H ₂ O Species Conservation
	Equations and Models	HTO Species Conservation
3 4		Leakage Model
		Permeations Model
	Port Information	1 Flow In
		2 Flow Out
		3 Tritium Leak and Permeation In
		4 Tritium Leak and Permeation Out

Table 4-10. Specification of PipeII-HTSE component block.

Table 4-11 Additional	innut narameters	of Pipe-HTSE of	component block
Table 4-11. Additional	i input parameters	S OI PIPE-TISE C	component block.

Input Parameter	Unit	Description
H ₂ O Initial Concentration	mole fraction	Average initial H ₂ O mole fraction in the pipe
HTO Initial Concentration	mole fraction	Average initial HTO mole fraction in the pipe

Table 4-12. Specification of Pipe-SI component block.

Icon	Specifications		
		HT Species Conservation	
		H ₂ Species Conservation	
		H ₂ O Species Conservation	
		HTO Species Conservation	
	Governing Equations and Models	H ₂ SO ₄ Species Conservation	
		HTSO ₄ Species Conservation	
Pipe - SI		HI Species Conservation	
		TI Species Conservation	
		Leakage Model	
		Permeations Model	
		1 Flow In	
	Port Information	2 Flow Out	

Icon	Specifications		
		HT Species Conservation	
		H ₂ Species Conservation	
		H ₂ O Species Conservation	
		HTO Species Conservation	
	Governing	H ₂ SO ₄ Species Conservation	
	Equations and Models	HTSO ₄ Species Conservation	
3 4		HI Species Conservation	
		TI Species Conservation	
		Leakage Model	
		Permeations Model	
	Port Information	1 Flow In	
		2 Flow Out	
		3 Tritium Leak and Permeation In	
		4 Tritium Leak and Permeation Out	

Table 4-13. Specification of PipeII-SI component block.

Table 4-14 Ad	Iditional innu	t narameters	of Pine-SI	component	block
1 auto 4-14. Au	unionai mpu	i parameters	or r ipe-or	component	DIOCK.

Input Parameter	Unit	Description		
H ₂ O Initial Concentration	mole fraction	Average initial H ₂ O mole fraction in the pipe		
HTO Initial Concentration	mole fraction	Average initial HTO mole fraction in the pipe		
H ₂ SO ₄ Initial Concentration	mole fraction	Average initial H ₂ SO ₄ mole fraction in the pipe		
HTSO ₄ Initial Concentration	mole fraction	Average initial HTSO ₄ mole fraction in the pipe		
HI Initial Concentration	mole fraction	Average initial HI mole fraction in the pipe		
HT Initial Concentration	mole fraction	Average initial HT mole fraction in the pipe		

Table 4-15. Specification of Isotope Exchange component block.

Icon	Specifications		
1 2 Isotope Exchange	Governing Equations and Models	$HT+H_2O = H_2+HTO$ Equilibrium	
		$HT+H_2SO_4 = H_2+HTSO_4$ Equilibrium	
		$HT+HI = H_2+TI$ Equilibrium	
	Port Information	1 Flow In	
		2 Flow Out	

Table 4-16. Input Parameters of Isotope I	Exchange Component Block.
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Input Parameter	Unit	Description
Isotope Exchange Model	#	Selection of Isotope Exchange Model for VHTR, HTSE, and SI Process

Table 4-17. Specification of Tritium External Source block.

Icon	Specifications		
Tritium External Source Governing Equations and Models		NONE	
	Port Information	Tritium External Source Out	

Table 4-18. Input parameters of Tritium External Source block.

Input Parameter	Unit	Description
External HT Source	m ³ (STP)/s	External HT Source into Pipe
External H ₂ Source	m ³ (STP)/s	External H ₂ Source into Pipe

4.4.3 Heat Exchanger Component

The **Heat Exchanger Component** category has two component blocks as shown in Table 4-2. Table 4-19 summarize the general **Heat Exchanger** component block which solves two species conservation for each side of the channel with the permeation model. On the other hand, Table 4-20 summarizes the **Heat Exchanger II** component model which solves only permeation model without species conservation in the component.

Icon	Specifications		
	Governing Equations and Models	HT Species Conservation (Primary)	
		H ₂ Species Conservation (Primary)	
		HT Species Conservation (Secondary)	
		H ₂ Species Conservation (Secondary)	
		Permeations Model	
Heat Exchanger	Port Information	1 Flow In (Primary)	
		2 Flow Out (Primary)	
		3 Flow In (Secondary)	
		4 Flow Out (Secondary)	

Table 4-19. Specification of Heat Exchanger component block.

Table A_20	Specification	of Heat Exchange	· II com	nonent block
1a010 + 20.	specification	Of ficat Exchange		ponent block.

Icon			Specifications
	Governing Equations and Models	Permeations Model	
1 Heat Exchanger II	Port Information	1	Flow In (Primary)
		2	Flow Out (Primary)
		3	Flow In (Secondary)
		4	Flow Out (Secondary)

4.4.4 Vessel Component

The Vessel Component category has three different types of vessel component blocks; VHTR, HTSE, and SI. Tables 4-21, 4-22, and 4-23 summarize the block specification s for Vessel-VHTR, Vessel-HTSE, and Vessel-SI blocks, respectively. The basic input parameters are the same as those of pipe components (See Tables 4-8, 4-11, 4-14).

Icon	Specifications			
	Governing Equations and Models	HT Species Conservation		
		H ₂ Species Conservation		
		Leakage Model		
3 2 4		Permeations Model		
1 > 2		1 Flow In		
Vessel - VHTR	Port Information	2 Flow Out		
		3 Tritium Leak and Permeation In		
		I Tritium Leak and Permeation Out		

Table 4-21. Specification of Vessel-VHTR component block.

Table 4-22. Specification of Vessel-HTSE component block.

Icon	Specifications		
	Governing Equations and Models	HT Species Conservation	
		H ₂ Species Conservation	
		H ₂ O Species Conservation	
		HTO Species Conservation	
3 4		Leakage Model	
1 > 2		Permeations Model	
Vessel - HTSE	Port Information	1 Flow In	
		2 Flow Out	
		3 Tritium Leak and Permeation In	
		Tritium Leak and Permeation Out	

Icon	Specifications		
		HT Species Conservation	
		H ₂ Species Conservation	
		H ₂ O Species Conservation	
		HTO Species Conservation	
	Governing	H ₂ SO ₄ Species Conservation	
	Equations and Models	HTSO ₄ Species Conservation	
3 4		HI Species Conservation	
		TI Species Conservation	
Vessel -SI		Leakage Model	
		Permeations Model	
	Port Information	1 Flow In	
		2 Flow Out	
		3 Tritium Leak and Permeation In	
		4 Tritium Leak and Permeation Out	

Table 4-23. Specification of Vessel-SI component block.

4.4.5 Containment Component

Table 4-24 summarizes the specifications of the **Containment** component block. This block is similar to the **VHTR-pipe** components and therefore the same input parameters are used here (See Table 4-8).

Table 4-24. Specification of Containment compone
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Icon	Specifications		
	Governing	HT Species Conservation	
		H ₂ Species Conservation	
	Equations and Models	Leakage Model	
	Wiedels	Permeations Model	
1 Containment	Port Information	1 Flow In	
		2 Flow Out	
		3 Tritium Leak and Permeation In	
		4 Tritium Leak and Permeation Out	

4.4.6 Purification System Component

Tables 4-25 and 4-26 summarize the specifications and the input parameters in the **Purification System** component block. This block does not solve any species conservation equations but solves the purification system model.

Table 4-25 S	necification	of Purification	System	component blo	ock
10010 + 20.0	peemeanon	of i unification	5 y Stern	component bio	JUR.

Icon			Specifications
1 👫 2	Governing Equations and Models	Puri	fication System Model
Purification Sustam		1	Flow In
Pumication System	Port Information	2	Flow Out

Table 4-26. Input parameters of Purification System component block.

Input Parameter	Unit	Description
Fraction of Mass Flow to Purification System	_	Fraction of flow to Purification System (0~1)
Fractional Efficiency of HT Purification		Efficiency of HT Purification (0~1)
Fractional Efficiency of H ₂ Purification		Efficiency of H_2 Purification (0~1)
Fractional Efficiency of H ₂ O Purification		Efficiency of H_2O Purification (0~1)
Fractional Efficiency of HTO Purification		Efficiency of HTO Purification (0~1)

4.4.7 Electrolyzer Component

Tables 4-27 and 4-28 summarize the specifications and the input parameters for the **Electrolyzer** component block. This block does not solve any species conservation equations, but solves the electrolyzer model that is based on the conversion ratio.

Table 4-27. Specification of Electrolyzer component block.

Icon	Specifications		
3 4	Governing Equations and Models	Electrolyzer Model	
1 Electrolyzer 2	Port Information	1 Flow In (Anode)	
		2 Flow Out (Anode)	
		3 Flow In (Cathode)	
		4 Flow Out (Cathode)	

Table 4-28. Input parameters of Electrolyzer component block.

Input Parameter	Unit	Description
System Conversion Ratio		H ₂ O conversion ratio by electrolysis (0~100%)

4.4.8 Flow Distribution Component

Tables 4-29, 4-30, and 4-31 summarize the **Flow Splitter** and **Flow Mixer** blocks which are related to the flow distributions. These blocks are not solving any physical equations or models, but are very important blocks when users develop complicated system configurations.
Table 4-29. Specification of Flow Splitter block.

Icon	Specifications		
. 2	Governing Equations and Models	NONE	
Flow Spliter	Port Information	1 Flow In	
		2 Flow Out-1	
		3 Flow Out-2	

Table 4-30. Input parameters of Flow Splitter block.

Input Parameter	Unit	Description
Separation Ratio	_	Separation Ratio of Flow Out-1 to Flow In (0~1)

Table 4-31. Specification of Flow Mixer block.

Icon	Specifications		
2	Governing Equations and Models	NONE	
Flow Mixer	Port Information	flow In-1	
		2 Flow In-2	
		3 Flow Out	

4.4.9 Boundary Flow Component

Tables 4-32 and 4-33 summarize the specifications and the input parameters of the **Boundary Flow** block which is used for providing flow sources in the open system (once-through).

Table 4-32.	Specification	of Boundary	Flow b	block.
	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~			

Icon	Specifications		
┣ 1	Governing Equations and Models	NONE	
boundary inflow	Port Information	1 Flow Out	

Input Parameter	Unit	Description
Flow Rate	m ³ (STP)/s	Flow rate of the boundary flow
Temperature	К	Average temperature of the boundary flow
Pressure	Ра	Average pressure of the boundary flow
HT Concentration	mole fraction	HT mole fraction of the boundary flow
H ₂ Concentration	mole fraction	H ₂ mole fraction of the boundary flow
H ₂ O Concentration	mole fraction	H ₂ O mole fraction of the boundary flow
HTO Concentration	mole fraction	HTO mole fraction of the boundary flow
H ₂ SO ₄ Concentration	mole fraction	H ₂ SO ₄ mole fraction of the boundary flow
HTSO ₄ Concentration	mole fraction	HTSO ₄ mole fraction of the boundary flow
HI Concentration	mole fraction	HI mole fraction of the boundary flow
HT Concentration	mole fraction	HT mole fraction of the boundary flow

Table 4-33. Input parameters of Boundary Flow component block.

4.4.10 Detector Component

Table 4-34 summarizes the specification of the **Detector** block which allows the user to access the flow or component information during or after simulation.

Table 4-34. Sp	ecification of	f Detector Block
----------------	----------------	------------------

Icon	Specifications		
1	Governing Equations and Models	NONE	
Detector	Port Information	1 Flow In (Signal)	

4.4.11 Data Save Component

Tables 4-35 and 4-36 summarize the specifications of the two data save blocks; **Save to File** and **Save to Workspace**. The **Save to File** block allows the user to save the flow conditions into file while the **Save to Workspace** block allows the user to save the file in the MATLAB Workspace.

Table 4-35. Specification of Save to File block.

Icon	Specifications		
↓untitled.mat Save to File	Governing Equations and Models	NONE	
	Port Information	flow In (Signal)	

Icon	Specifications		
1 > simout	Governing Equations and Models	NONE	
Save to Workspace	Port Information	1 Flow In (Signal)	

Table 4-36. Specification of Save to Workspace Block.

4.5 Solver Setting

The TPAC is using the solver embedded in the MATLAB SIMULINK. The Solver Setting window can be opened by selecting **Simulation/Configuration Parameters...** in the Main Menu (See Figure 4-34).

🚺 untitled *	property and the second			
File Edit View	Simulation Format Tool	; Help		
	Start Stop	Ctrl+T	▾ 🛱 🛗 🔮 🛗	🖡 🗟 🗊 🕆
	Configuration Paramet	ers Ctrl+E		
	✓ Normal Accelerator External			
*				
Show the active cor	nfiguration parameters dialog	100%	ode45	

Figure 4-34. Solver Setting menu.

The following is the description about the Solver Setting Panel documented in the MATLAB SIMULINK user manual (Mathworks 2009). For more details about the solver setting, the MATLAB user manual is recommended.

The Solver configuration parameters pane allows the user to specify a simulation start and stop time and select and configure a solver for a particular simulation configuration (See Figure 4-35).

Simulation time			
Start time: 0.0		Stop time: 20	
Solver options			
Туре:	Variable-step 🗾	Solver:	ode45 (Dormand-Prince) 🗾
Max step size:	auto	Relative tolerance:	1e-3
Min step size:	auto	Absolute tolerance:	1e-6
Initial step size:	auto		
Zero crossing control:	Use local settings 📃 💌		

Figure 4-35. Solver Panel.

The Solver pane contains the following control groups:

- *Simulation time*. This control group enables us to specify the simulation start and stop time. It contains the following controls.
 - **Start time**. Specifies the simulation start time. The default start time is 0.0 seconds.
 - **Stop time**. Specifies the simulation stop time. The default stop time is 10.0 seconds. Specify "inf" to cause the simulation to run until users pause or stop it.

Simulation time and actual clock time are not the same. For example, running a simulation for 10 seconds usually does not take 10 seconds. The amount of time it takes to run a simulation depends on many factors, including the model's complexity, the solver's step sizes, and the computer's speed.

• *Solver Options*. The Solver options controls group allows us to specify the type of solver to be used and simulation options specific to that solver (See Figure 4-36).

Solver options			
Туре:	Variable-step 🗾	Solver:	ode45 (Dormand-Prince) 📃
Max step size:	auto	Relative tolerance:	1e-3
Min step size:	auto	Absolute tolerance:	1e-6
Initial step size:	auto		
Zero crossing control:	Use local settings 📃 💌		

Figure 4-36. Solver Options Control.

The contents of the group depend on the solver type.

- General Solver Options. The follow options always appear:
 - **Type.** Specifies the type of solver to be used to solve the currently selected model, either Fixed-step or Variable-step.
 - **Solver.** Specifies the solver used to simulate this configuration of the current model. The associated pull-down list displays available solvers of the type specified by the Type control. To specify another solver of the specified type, select the solver from the pull-down list.

The other controls that appear in this group depend on the type of solver users have selected.

• *Variable-Step Continuous Solver Options*. The following options appear when users select any of the SIMULINK variable-step continuous solvers.

Γ	Solver options			
	Туре:	Variable-step 🗾	Solver:	ode45 (Dormand-Prince) 🗾
	Max step size:	auto	Relative tolerance:	1e-3
	Min step size:	auto	Absolute tolerance:	1e-6
	Initial step size:	auto		
	Zero crossing control:	Use local settings 📃 💌	[

Figure 4-37. Variable-Step Continuous Solver Options.

- **Max step size.** Specifies the largest time step the solver can take. The default is determined from the start and stop times. If the stop time equals the start time or is inf, SIMULINK chooses 0.2 sec. as the maximum step size. Otherwise, it sets the maximum step size to

$$h_{\max} = \frac{t_{stop} - t_{start}}{50} \tag{4-1}$$

Generally, the default maximum step size is sufficient. If users are concerned about the solver missing a significant behavior, change the parameter to prevent the solver from taking too large a step. If the time span of the simulation is very long, the default step size might be too large for the solver to find the solution. Also, if our model contains periodic or nearly periodic behavior and users know the period, set the maximum step size to some fraction (such as 1/4) of that period.

- **Initial step size.** By default, the solver selects an initial step size by examining the derivatives of the states at the start time. If the first step size is too large, the solver might step over important behavior. The initial step size parameter is a suggested first step size. The solver tries this step size but reduces it if error criteria are not satisfied.
- Min step size. This option appears only for variable-step continuous solvers. Specifies the smallest time step the selected variable-step solver can take. If the solver needs to take a smaller step to meet error tolerances, it issues a warning indicating the current effective relative tolerance. This parameter can be either a real number greater than zero or a two-element vector where the first element is the minimum step size and the second element is the maximum number of minimum step size warnings to be issued before issuing an error. Setting the second element to zero results in an error the first time the solver must take a step smaller than the specified minimum. This is equivalent to changing the minimum step size violation diagnostic to error on the Diagnostics pane. Setting the second element to -1 results in an unlimited number of warnings. This is also the default if the input is a scalar. The default values for this parameter are a minimum step size on the order of machine precision and an unlimited number of warnings.
- Relative tolerance. Relative tolerance for this solver.
- Absolute tolerance. Absolute tolerance for this solver.
- **Maximum order.** This option appears only if users select the ode15s solver, which is based on numerical differentiation formulas (NDFs) of orders 1–5. Although the higher order formulas are more accurate, they are less stable. If our model is stiff and requires more stability, reduce the maximum order to 2 (the highest order for which the NDF is A-stable). As an alternative, users can try using the ode23s solver, which is a lower order (and A-stable) solver.

5. VERIFICATION OF THE CODE

TPAC is a dynamic simulation code developed at INL for tritium behaviors analyses in the VHTR system (named as TPAC) as described in detail in the previous sections. The TPAC was designed to solve (1) mass conservations of tritium and hydrogen, (2) core tritium generation, (3) tritium permeation through the pipe walls or heat exchanger walls, (4) leakage, (5) purification systems, (6) isotope-exchange, and (7) electrolyzer.

To achieve code reliability, the TPAC was verified based on analytical solutions and experimental data. The main focus of verification work was to determine if:

- The equations and models are correctly implemented into the code to represents the phenomena of interest.
- There is any bug in the code.
- The code needs any improvements or modifications.

The Peach Bottom data and analytical solutions were used as the reference data for the verification. The Peach Bottom reactor was a helium-cooled graphite-moderated reactor built in the United States in 1958. Figure 5-1 shows the schematics of the Peach Bottom Reactor. The primary circuit consisted of two loops, each containing a helium compressor and steam generator. The total helium flow of 210,000 kg/h was divided equally between the two loops. Coolant temperatures at the core inlet and outlet of the reactor vessel were 345 and 714°C, respectively, and the primary loop pressure was approximately 2.4 MPa (335 psig). The reactor and steam generator were connected by a concentric duct. Coolant exiting the reactor flows in the inner pipe of the concentric duct. The steam generators were forced recirculation, drum type boilers that have pendant U-tube economizer, evaporator, and super-heater sections.

In the process of verifying the Peach Bottom reactor, Core 2 operation data was evaluated by Wichner and Dyer (1979). Core 1 fuel particles were coated with a single layer of pyrolytic graphite, solely to prevent hydrolysis during manufacture. The Core 2 fuel particles were designed to be BISO (fuel coated with two ceramic layers) in order to retain gaseous fission products during the life (from July 1970 to October 1974).



Figure 5-1. Primary coolant system of the Peach Bottom reactor (Wichner and Dyer 1979).

Analytical solutions and the THYTAN code results, previously verified in FY-07 and FY-08, were also used in verifying the TPAC. The following models were taken into considerations in the verification process:

- Mass conservation equations (governing equations).
- Reactor core model for tritium generation.
- Tritium and hydrogen permeation model through the metal.
- Leak model of the reactor system.
- Purification system model.
- System coupling and integration model.

5.1 Verification of Core Model

The core model for the tritium birth was verified first. The core model involves various tritium birth models from different sources, and therefore it generally affects total tritium distributions and behaviors the most. The core model in the TPAC consists of the following tritium sources:

- Tritium birth from ternary fission
- Tritium birth from neutron reaction with lithium (⁶Li, ⁷Li)
- Tritium birth from neutron reaction with ³He in the coolant
- Tritium birth from neutron reaction with ¹⁰B.

5.1.1 Tritium Birth from Ternary Fission

Figure 5-2 shows the tritium birth modeling of the TPAC for Ternary fission. Table 5-1 summarizes the input parameters used for verification for this ternary fission model. In the ternary fission block, the following tritium source equation has been implemented.

$$\frac{d(N_{T(Ter)})}{dt} = K \cdot P \cdot Y - \lambda \cdot N_{T(Ter)}$$
(5-1)

where

 $N_{T(Ter)}$ = number of tritium atoms from ternary fission

K = fission rate per thermal megawatt [fission/MW/s]

P = reactor power [MW]

Y = average yield per fission [1/fission]

$$\lambda$$
 = tritium decay constant [1/s].



Figure 5-2. TPAC tritium birth modeling from ternary fission.

Table 5-1. TPAC Input parameters for calculation of tritium birth from ternary fissi
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Input Parameters	Values
Reactor Power (MWth)	66.55
Fission Rate per Thermal Power (Fission/MWth/s)	3.12e16
Average Fission per Fission (1/s)	1e-4
Initial Amount of Tritium (#)	0
Calculation Time (s)	1.34e8

The following analytical solution was used for validation of the ternary fission model in the TPAC.

$$N_T(t_f) = K \cdot \overline{P} \cdot Y\left(\frac{1 - \exp(-\lambda \cdot t_f)}{\lambda}\right)$$
(5-2)

where

 $N_T(t_f) = \text{ atoms of tritium at time } t_f$

K = fission rate per thermal megawatt (3.121 × 10¹⁶ fissions/MW/s)

Y = average yield per fission (1×10^{-4})

 λ = tritium decay constant (1.793 × 10⁻¹⁹ s⁻¹).

An average power, \overline{P} was calculated using the following equation.

$$\overline{P} = P_{rated} \left(EFPD / t_f \right)$$
(5-3)

where

 P_{rated} = rated power (115 MWth)

EFP = equivalent full power days at EOL (897 days)

 t_f = duration of Core 2 operation, including shutdown (1.34 × 10⁸ s or 1,550 days).

Table 5-2 summarizes the computed and reported results. The reported tritium birth by ternary fission during 1,550 days was 1,210 Ci (= 4.43e13 Bq). The TPAC results show very good agreement with the reported value and the analytical solution. They are also in good agreement with the THYTAN solution.

Table 5-2. Comparison of tritium activity by ternary fission.

			Activity	
Reported V Bo (Wichner a	Value - Peach ottom nd Dyer 1979)	Analytical Solution	Computed Solution from THYTAN (Ohashi and Sherman 2007)	Computed Solution by TPAC
(Ci)	(Bq)	(Bq)	(Bq)	(Bq)
1,210	4.43×10^{13}	4.43×10^{13}	4.42×10^{13}	4.436×10^{13}

5.1.2 Tritium Birth from Li Impurities

Figure 5-3 shows the TPAC tritium birth modeling from Li impurities. For verification, the tritium birth from the Li has been converted to radioactivity.



Figure 5-3. TPAC tritium birth modeling for neutron reaction with Li impurities.

The TPAC tritium source block from Li impurities, which was specified as 'Birth from 6Li' in Figure 5-3, was designed to solve the following equations:

$$\frac{d(N_{Li7})}{dt} = -\phi_f \cdot \sigma_{Li7H3} \cdot N_{Li7}$$
(5-4)

$$\frac{d(N_{T(Li7)})}{dt} = \phi_f \cdot \sigma_{Li7T} \cdot N_{Li7} - \lambda \cdot N_{T(Li7)}$$
(5-5)

where

 N_{Li7} = number of ⁷Li atoms, excluding ¹⁰B source

 $N_{T(Li7)}$ = number of tritium atoms from ⁷Li, excluding birth from ¹⁰B

 ϕ_f = fast neutron flux [neutrons/cm²/s]

 σ_{Li7T} = effective cross section for ⁷Li (n, n\alpha) ³H [cm²].

Table 5-3 summarizes the input parameters used for verification of the ⁶Li neutron reaction model in the TPAC.

Component	Thermal Neutron Flux (neutrons/cm ² /s)	Initial Amount of ⁶ Li (moles)
Sleeve	2.82×10^{13}	6.99×10^{-4}
Spine	2.82×10^{13}	5.17×10^{-5}
Removal radial reflector	Core 1 : 2.57×10^{13} Core 2 : 2.82×10^{13}	7.56×10^{-4}
Permanent radial reflector	Core 1 : 1.28×10^{13} Core 2 : 1.41×10^{13}	1.12×10^{-3}
Axial reflector	1.41×10^{13}	$6.74 imes 10^{-4}$
Fuel matrix	2.82×10^{13}	7.76×10^{-4}

Table 5-3. Input parameters for validation of tritium birth model from ⁶Li in the TPAC.

The reference birth of tritium from ⁶Li for verification was estimated by the following analytical solution:

$$N_{T}(t_{f}) = \left(\frac{\phi_{th} \cdot \sigma_{Li6T} \cdot N_{6}(0)}{\lambda - \phi_{th} \cdot \sigma_{Li6T}}\right) \cdot \left[\exp(-\phi_{th} \cdot \sigma_{Li6T} \cdot t_{f}) - \exp(-\lambda \cdot t_{f})\right]$$
(5-6)

where

 $N_T(t_f) =$ atoms of tritium from ⁶Li at time t_f

 σ_{Li6T} = effective cross section for ⁶Li (n, α) T (4.08 × 10⁻²² cm²)

 $N_6(0)$ = initial amount of ⁶Li atoms.

Table 5-4 summarizes the reported and computed tritium activity. As reported by Ohashi and Sherman (2007), there were some unknown discrepancies between reported values and analytical solutions that have not yet been revealed. However, their THYTAN solutions showed good agreement with the analytical solutions. The TPAC also showed very good agreement with the analytical solutions and THYTAN results for various regions of the core. It verifies that the TPAC solves Equations (5-4) and (5-5) correctly.

		Activity			
Graphite	Reported Value (Peach Bottom)		Analytical Solution	THYTAN (Ohashi and Sherman 2007)	TPAC Solution
Component	(Ci)	(Bq)	(Bq)	(Bq)	(Bq)
Sleeve	14.0	5.13×10^{11}	5.12×10^{11}	5.12×10^{11}	5.12×10^{11}
Spine	1.0	3.66×10^{10}	$3.78 imes 10^{10}$	3.78×10^{10}	3.78×10^{11}
Removal radial reflector	16.4	6.01×10^{11}	5.76×10^{11}	5.75×10^{11}	5.76×10^{11}
Permanent radial reflector	18.8	6.89×10^{11}	6.72×10^{11}	6.69×10^{11}	6.68×10^{11}
Axial reflector	9.2	3.37×10^{11}	3.42×10^{11}	3.42×10^{11}	3.42×10^{11}
Fuel matrix	13.1	4.80×10^{11}	5.68×10^{11}	5.68×10^{11}	5.68×10^{11}

Table 5-4. Comparison of tritium activity from ⁶Li at the Core 2 operation of the Peach Bottom reactor.

5.1.3 Tritium Birth from ³He Impurities in the Coolant

Figure 5-4 shows the TPAC tritium birth modeling for neutron reaction of ³He impurities in the coolant. The tritium source block by ³He neutron reaction was designed to solve the following equations:

$$\frac{d(N_{He3})}{dt} = f \cdot N_{He3}^{\circ} - f \cdot N_{He3} - \phi_{He} \cdot \sigma_{He3T} \cdot N_{He3}$$
(5-7)

$$\frac{d(N_{T(He3)})}{dt} = \phi_{He} \cdot \sigma_{He3T} \cdot N_{He3} - \lambda \cdot N_{T(He3)}$$
(5-8)

$$\phi_{He} = \frac{W_{core}}{W_{total}} \cdot \phi_{th} \tag{5-9}$$

where

 N_{He3} = number of ³He atoms $N_{T(He3)}$ = number of tritium atoms from ³He f = fractional supply rate of helium coolant [1/s]

 N_{He3}° = number of ³He atoms in the supply helium

 σ_{He3T} = effective cross section for ³He (n, p) T [cm²]

 ϕ_{He} = average thermal neutron flux experienced by the total primary helium inventory [n/cm²/s]

 W_{core} = helium inventory in core [kg]

 W_{total} = total primary helium inventory [kg].



Figure 5-4. TPAC Tritium birth model for neutron reaction with ³He impurities.

For obtaining the reference data, the birth of tritium from ³He was estimated by the following analytical equations (Wichner and Dyer [1979]).

$$N_{T1} = \frac{\left(V_1 + V_7\right) \cdot \phi_{th} \cdot \sigma_{He3T} \cdot N_3}{Q + \lambda \cdot V}$$
(5-10)

for the sleeve graphite

$$N_{T2}(t_f) = \frac{V_3}{V} \cdot \phi_{th} \cdot \sigma_{He3T} \cdot N_3 \left[\frac{1 - \exp(-\lambda \cdot t_f)}{\lambda} \right],$$
(5-11)

for the removal radial reflector

$$N_{T3}(t_f) = \frac{V_4}{V} \cdot \phi_{th} \cdot \sigma_{He3T} \cdot N_3 \left[\frac{1 - \exp(-\lambda \cdot t_f)}{\lambda} \right],$$
(5-12)

for the permanent radial reflector

$$N_{T4}(t_f) = \frac{V_6}{V} \cdot \phi_{th,3} \cdot \sigma_{He3T} \cdot N_3 \left[\frac{1 - \exp(-\lambda \cdot t_f)}{\lambda} \right],$$
(5-13)

and for the axial reflector

$$N_{T5}(t_f) = \frac{V_5}{V} \cdot \phi_{th,2} \cdot \sigma_{He3T} \cdot N_3 \left[\frac{1 - \exp(-\lambda \cdot t_f)}{\lambda} \right]$$
(5-14)

where

N_{T1}	=	total moles of tritium circulating in the reactor with the coolant
N_{T2}	=	total moles of tritium born in the sleeve graphite from ³ He
N_{T3}	=	total moles of tritium born in the removal radial reflector from ³ He
N_{T4}	=	total moles of tritium born in the permanent reflector from ³ He
N_{T5}	=	total moles of tritium born in the axial reflector from ³ He
V_1	=	volume of coolant passage in core $(1.77 \times 10^6 \text{ cm}^3)$
V_2	=	ex-core primary system volume $(1.88 \times 10^8 \text{ cm}^3)$
V_3	=	connected porosity in sleeve graphite $(3.44 \times 10^5 \text{ cm}^3)$
V_4	=	connected porosity in removal radial reflector ($8.46 \times 10^5 \text{ cm}^3$)
V_5	=	connected porosity in axial reflector $(3.92 \times 10^5 \text{ cm}^3)$
V_6	=	connected porosity in permanent radial reflector $(2.32 \times 10^6 \text{ cm}^3)$
V_7	=	purge flow volume within the fuel elements $(6.32 \times 10^5 \text{ cm}^3)$
V	=	effective helium volume of the primary system $(V_1 + V_2 + V_3 + V_4 + V_5 + V_6 \ [1.94 \times 10^8 \text{ cm}^3])$
ϕ_{th}	=	average thermal neutron flux in core and removal radial reflector, Core 2 $(2.82 \times 10^{13} \text{ neutrons/cm}^2/\text{s})$
$\phi_{th,2}$	=	average thermal neutron flux in axial reflector $(1.41 \times 10^{13} \text{ neutrons/cm}^2/\text{s})$
$\phi_{th,3}$	=	average thermal neutron flux in permanent radial reflector $(1.41 \times 10^{13} \text{ neutrons/cm}^2/\text{s})$
σ_{He3T}	=	effective cross section for ³ He (n, p) T ($2.28 \times 10^{-21} \text{ cm}^2$)
Q	=	flow to chemical cleanup system plus 10% of fuel element purge flow ($Q_1(2.40 \times 10^4 \text{ cm}^3/\text{s})$) + leakage flow rate from primary system ($Q_2[25.5 \text{ cm}^3/\text{s}]$).

The total moles of ³He in the primary system, N_3 , is governed by the relation in the following equation:

$$\frac{dN_3}{dt} = -\sigma_{He3T} \cdot N_3 \left\{ \left(\frac{V_1 + V_3 + V_4 + V_7}{V} \right) \cdot \phi_{th} + \left(\frac{V_5 + V_6}{V} \right) \cdot \phi_{th,2} \right\} + Q_2 \left([^3He]_i - [^3He] \right)$$
(5-15)

where

- $\begin{bmatrix} {}^{3}He \end{bmatrix}_{i} = {}^{3}\text{He concentration in makeup helium (5.78 × 10⁻¹¹ moles/cm³ based on 0.16 ppm ³He in helium)}$
- $\begin{bmatrix} {}^{3}He \end{bmatrix} = {}^{3}$ He concentration in primary system (= N_{3} / V).

Table 5-5 summarizes the calculated tritium birth from ³He in the Core 2 operation in the Peach Bottom reactor. The input parameters are summarized in Tables 5-6 and 5-7. Ohashi and Sherman (2007) reported that the values are larger than both analytical and computed solutions within 10–60%. However, the reason is still unknown. The TPAC solutions show good agreement with analytical solutions and THYTAN results. It verifies that the TPAC solves Equations (5-7), (5-8), and (5-9), correctly.

	Activity				
	Reported Value (Peach Bottom)		Analytical Solution	THYTAN (Ohashi and Sherman 2007)	TPAC Solution
Region	(Ci)	(Bq)	(Bq)	(Bq)	(Bq)
In sleeve graphite	5.4	1.98×10^{11}	1.31×10^{11}	1.26×10^{11}	1.28×10^{11}
In removable radial reflector	13.5	4.95×10^{11}	3.20×10^{11}	3.09×10^{11}	3.12×10^{11}
In permanent radial reflector	15.5	5.68×10^{11}	5.43×10^{11}	5.24×10^{11}	5.31×10^{11}
In axial reflector	3.1	1.14×10^{12}	9.17×10^{11}	8.86×10^{11}	8.98×10^{11}

Table 5-5. Comparison of tritium activity from ³He at the Core 2 operation of the Peach Bottom HTGR.

Table 5-6. Input data for calculation of tritium birth from ³He (Ohashi and Sherman 2007).

Parameter	Unit	Value
Effective cross section for ³ He (n, p) T	cm	2.28×10^{-21}
³ He concentration in makeup helium	ppm	0.16
Helium inventory in primary system	kg	269

Table 5-7. Input data for the calculation of tritium birth from ³He (Ohashi and Sherman 2007).

Subnode	Thermal Neutron Flux (neutrons/cm ² /s)	Helium Inventory (kg)
In circulating coolant	2.82×10^{13}	3.33
In sleeve graphite	2.82×10^{13}	0.477
In removal radial reflector	2.82×10^{13}	1.17
In permanent radial reflector	1.41×10^{13}	3.22
In axial reflector	1.41×10^{13}	0.543

5.1.4 Tritium Birth from ¹⁰B Impurities in the Coolant

Figure 5-5 shows the TPAC tritium birth modeling for neutron reaction of ¹⁰B impurities in the coolant. The tritium source block by ¹⁰B neutron reaction was designed to solve the following equations.

$$\frac{d(N_{B10})}{dt} = -\left(\phi_{th} \cdot \sigma_{B10Li7} + \phi_f \cdot \sigma_{B10T}\right) \cdot N_{B10}$$
(5-16)

$$\frac{d(N_{Li7(B10)})}{dt} = \phi_{th} \cdot \sigma_{B10Li7} \cdot N_{B10} - \phi_f \cdot \sigma_{Li7T} \cdot N_{Li7(B10)}$$
(5-17)

$$\frac{d(N_{T(B10)})}{dt} = \phi_f \cdot \sigma_{Li7T} \cdot N_{Li7(B10)} + \phi_f \cdot \sigma_{B10T} \cdot N_{B10} - \lambda \cdot N_{T(B10)}$$
(5-18)

where

N_{B10}	=	number of ¹⁰ B atoms
$N_{Li7(B10)}$	=	number of ⁷ Li atoms from ¹⁰ B
$N_{T(B10)}$	=	number of tritium from ¹⁰ B
$\sigma_{{\scriptscriptstyle B10Li7}}$	=	effective cross section for $^{10}B(n, \alpha)$ ⁷ Li [cm ²]
~		10 0 0

 σ_{B10T} = effective cross section for ¹⁰B (n, 2 α) ³H [cm²].



Figure 5-5. TPAC Tritium birth model for neutron reaction with ¹⁰B impurities.

For the reference data, the tritium birth from ¹⁰B was evaluated by the following equation:

$$\frac{N_{T}(t_{f})}{N_{10}(0)} = \frac{b \cdot a}{(b-a) \cdot (\lambda - a)} \left(\exp(-a \cdot t_{f}) - \exp(-\lambda \cdot t_{f}) \right) - \frac{b \cdot a}{(b-a) \cdot (\lambda - b)} \left(\exp(-b \cdot t_{f}) - \exp(-\lambda \cdot t_{f}) \right) + \frac{c}{(\lambda - a)} \left(\exp(-a \cdot t_{f}) - \exp(-\lambda \cdot t_{f}) \right)$$
(5-19)

where

 $a = \beta \phi th \sigma B^{10} Li^7$ $b = \phi f \sigma Li^7 T$ $c = \phi f \sigma B^{10} T$ $\beta = \text{self-shielding factor (0.0141)}$ $\phi_{th} = \text{average thermal neutron flux (2.82 \times 1,013 \text{ neutrons/cm}^2/\text{s})}$

 ϕ_f = average fast neutron flux (2.26 × 1,013 neutrons/cm²/s) σ_{B10Li7} = effective cross section for ¹⁰B (n, α) ⁷Li (1.63 × 10⁻²¹ cm²) σ_{Li7T} = effective cross section for ⁷Li (n, n α) ³H (1.53 × 10⁻²⁵ cm²) σ_{B10T} = effective cross section for ¹⁰B (n, 2 α) ³H (5.00 × 10⁻²⁶ cm²) $N_{10}(0)$ = initial amount of ¹⁰B atoms (20.15 moles).

Tritium from ¹⁰B is produced in the poisoned spine and in the control rod. However, because the boron level and control rod position varies with time in the control rod, only the poisoned spine was taken into consideration here. Table 5-8 shows the calculation result and reported values. Computed values by the TPAC show good agreement with the analytical solutions.

Table 5-8. Comparison of tritium activity from ¹⁰B in the poisoned spine at the Core 2 operation of Peach Bottom reactor.

		Activity		
Report (Peach	ed Value Bottom)	Analytical Solution	THYTAN (Ohashi and Sherman 2007)	TPAC Solution
(Ci)	(Bq)	(Bq)	(Bq)	(Bq)
85.7	3.14×10^{12}	3.19×10^{12}	3.18×10^{12}	3.19×10^{12}

5.2 Verification of Permeation Model

In this section, the TPAC tritium permeation model has been verified. Figure 5-6 shows two modeling for verification. Figure 5-6 (a) is the modeling of tritium permeation through the vessel or pipe walls out of the components. Figure 5-6 (b) is the modeling of tritium permeation through the heat exchanger walls.



(a) Tritium permeation modeling in the vessel or pipe walls to outside.



(b) Tritium permeation modeling in the heat exchanger walls.

Figure 5-6. Tritium permeation modeling by INL Tritium code.

Two species permeations in the TPAC were taken into consideration; (1) hydrogen and (2) tritium. The TPAC was designed to solve the forms of general tritium permeation equations described below.

The permeation rate of H₂ at the heat exchanger, $R_{HX,H2}$ [m³ (STP)/s], and at the co-axial pipe, $R_{copipe,H2}$ [m³ (STP)/s], is estimated using

$$R_{HX,H2,j}\left(or \quad R_{copipe,H2,j}\right) = \frac{A}{l} \cdot k_{p,H} \cdot \left(\sqrt{P_{H2,h}} - \sqrt{P_{H2,l}}\right)$$
(5-20)

where

A = heat transfer area or surface area [m²]

l = thickness of heat transfer tube or component casing [m]

 $k_{p,H}$ = permeability of hydrogen [m³ (STP)/m/s/Pa^{0.5}]

- $P_{H2,h}$ = partial pressure of H₂ at high pressure side [Pa]
- $P_{H2,l}$ = partial pressure of H₂ at low pressure side [Pa].

l is calculated by

$$l = r_o \cdot \ln\left(\frac{r_o}{r_i}\right) \tag{5-21}$$

where

 r_o = outer radius of heat transfer tube [m]

 r_i = inner radius of heat transfer tube [m].

Permeability is calculated by using the Arrhenius equation:

$$k_p = f \cdot \exp\left(\frac{-E}{RT}\right) \tag{5-22}$$

where

f = pre-exponential factor of permeability [m³ (STP)/m/s/Pa^{0.5}]

- E = activation energy [J/mol]
- R = ideal gas constant [J/mol-K]
- T = temperature [K].

The permeation rate of H₂ though the outer wall of the component and piping, $R_{comp,H2}$, is calculated by excluding the H₂ partial pressure at the low pressure side as follows:

$$R_{comp,H2,j} = \frac{A}{l} \cdot k_{p,H} \cdot \sqrt{P_{H2,h}} \quad .$$
(5-23)

The permeation rate of HT at the heat exchanger, $R_{HX,H2}$ [m³ (STP)/s], and at the co-axial pipe, $R_{copipe,H2}$ [m³ (STP)/s], is estimated by considering the effect of the existence of hydrogen on the adsorption-dissociation and recombination-desorption step as follows:

$$R_{HX,HT,j}\left(or \quad R_{copipe,HT,j}\right) = \frac{A}{l} \cdot k_{p,T} \cdot \left(1 - \frac{P_{H2,l}}{P_{Total,l}}\right) \left(\frac{P_{HT,h}}{\sqrt{P_{H2,h} + P_{HT,h}}} - \frac{P_{HT,l}}{\sqrt{P_{H2,l} + P_{HT,l}}}\right)$$
(5-24)

where

 $k_{p,T}$ = permeability of tritium [m³ (STP)/m/s/Pa^{0.5}]

 $P_{HT,h}$ = partial pressure of HT at high pressure side [Pa]

 $P_{HT,l}$ = partial pressure of HT at low pressure side [Pa]

 $P_{Total,l}$ = total pressure at low pressure side [Pa].

The permeation rate of HT through the outer wall of the component and piping, $R_{comp,HT}$, is calculated by using the following equation:

$$R_{comp,HT,j} = \frac{A}{l} \cdot k_{p,T} \cdot \frac{P_{HT,h}}{\sqrt{P_{H2,h} + P_{HT,h}}}$$
(5-25)

For verification of tritium permeation models, the permeation rates of the heat transfer tubing obtained from the Peach Bottom HTGR steam generator at the end of life (Yang, Baugh, and Baldwin 1977) has been selected to be used. The reported permeation rate, J, of the Incoloy 800 super-heater tubing, steel evaporator tubing, and low-carbon steel economizer are as follows:

for the economizer

$$J = 7.78 \times 10^{1} \cdot \exp(-4500/T) (493 \text{ K} - 623 \text{ K}) [\mu \text{Ci/m}^{2}/\text{h}]$$
(5-26)

for the evaporator

$$J = 6.97 \times 10^3 \cdot \exp(-6830/T) (573 \text{ K}-693 \text{ K}) [\mu \text{Ci/m}^2/\text{h}]$$
(5-27)

for the super-heater

$$J = 1.72 \times 10^3 \cdot \exp(-6440/T) \quad (673 \text{ K} - 973 \text{ K}) \left[\mu \text{Ci/m}^2/\text{h}\right]. \tag{5-28}$$

Calculation of the tritium permeation rate for the super-heater, evaporator, and economizer was carried out. Table 5-9 shows the size and dimension of the heat exchangers. The experimental conditions—tritium concentration in the feed side of 0.461 ppb, feed helium pressure of 0.1013×10^5 Pa, and the permeation chamber pressure of 1.33×10^{-3} Pa—were employed as boundary conditions. The effect of impurity hydrogen was ignored. The hydrogen concentration of 199 ppm was assumed for the boundary condition for the primary side.

Sample	Effective Area (m ²)	Outer Radius (m)	Inner Radius (m)
Economizer (inlet)	103.47×10^{-4}	6.350×10^{-3}	4.775×10^{-3}
Evaporator (inlet)	68.34×10^{-4}	9.525×10^{-3}	7.300×10^{-3}
Super-heater (inlet)	90.16×10^{-4}	9.525×10^{-3}	6.350×10^{-3}

Table 5-9. The heat transfer tube dimensions of the steam generator in the Peach Bottom reactor (Ohashi and Sherman 2007).

Table 5-10 summarizes the analytical solutions of Equations (5-26) through (5-28), THYTAN results (Ohashi and Sherman 2007), and the TPAC solutions. In this table, the TPAC solutions are in good agreement with analytical solutions and reported THYTAN results.

			Computed	Solution
		Analytical Solutions	THYTAN (Ohashi and Sherman 2007)	TPAC Solution
	Temp. (K)	Permeation Rate (μCi/m ² /h)	Permeation rate (μCi/m²/h)	Permeation rate (μCi/m ² /h)
~ .	673	1.202×10^{-1}	1.202×10^{-1}	1.21×10^{-1}
Super-heater	823	6.873×10^{-1}	6.876×10^{-1}	6.91×10^{-1}
(iniet)	973	2.298	2.297	2.309
	573	4.640×10^{-2}	4.644×10^{-2}	4.639×10^{-2}
Evaporator	633	1.436×10^{-1}	1.437×10^{-1}	1.436×10^{-1}
(iniet)	693	3.656×10^{-1}	3.658×10^{-1}	3.655×10^{-1}
	493	8.450×10^{-3}	8.454×10^{-3}	8.495×10^{-3}
Economizer	558	2.447×10^{-2}	2.448×10^{-2}	2.459×10^{-2}
(inlet)	623	5.676×10^{-2}	5.678×10^{-2}	5.701×10^{-2}

Table 5-10. Comparison of permeation rates.

5.3 Verification of Leak Model

The verification of the leak model in the TPAC is described in this section. Figure 5-7 shows two different modeling for verification. In Figure 5-7(a), the primary and the containment were considered to be single components. To maintain the primary tritium concentration constant, the same amount of leakage from the primary side has been re-injected back to the primary side. In Figure 5-7(b), the primary side and the secondary side have been modeled to be two different loops with convective flows. In this model, leakage and mass conservation are verified together.

The TPAC input data is summarized in Table 5-11. Since the information on the containment vessel temperature is not available, the average containment vessel temperature is assumed to be 323 K. The tritium concentration in the primary loop of $2 \times 10^{5} \,\mu\text{Ci/cm}^{3}$ corresponds to 1.98×10^{3} ppb under the reported average temperature of 809 K and a pressure of 23 atm in the primary coolant. In order to keep the tritium concentration in the primary coolant 1.98×10^{3} ppb, the tritium release rate to the primary coolant was adjusted by the core model. The initial concentration of tritium in the containment vessel was set to zero.



(a) Tritium leakage modeling for single primary and single containment system.



(b) Tritium leakage modeling for duel primary and dual containment system. Figure 5-7. Tritium leak modeling by INL Tritium code.

Parameter	Unit	Value
Leak rate from primary to containment vessel	1/h	4.15×10^{-4} a
Primary loop volume	m ³	232
Primary loop pressure	Ра	$2.33 \times 10^{6 \text{ b}}$
Primary loop temperature	К	809
Leak rate from containment vessel to atmosphere	1/h	4.17×10^{-4} c
Containment vessel volume	m ³	$1.56 imes 10^{4 d}$
Containment vessel pressure	Ра	$1.57 \times 10^{5 e}$
Containment vessel temperature	К	323 ^f
a. Based on 3.2 kg/day. b. Based on 23 atm. c. Based on 0.1%/day. d. Based on a nitrogen volume of 2.04 × 104 m ³ at a pressure of e. Based on 8 psig. f Assumed	1.57×10^5 Pa and as	sumed temperature of 323 K.

Table 5-11. TPAC input data for verification of the leak model.

In order to verify the TPAC leak model, the tritium concentration in the containment vessel was estimated using the reported leak rate and the tritium concentration in the primary loop as Ohashi and Sherman (2007) did by using the THYTAN code. The TPAC solutions for the tritium concentration in the containment vessel were compared with the experimental result, the reported analytical solution, and THYTAN results.

The containment vessel's deinerting operation was carried out approximately every 127 days. The reported analytic solution for the tritium concentration in the containment vessel at the end of 127 days was $2.1 \times 10^{-7} \,\mu\text{Ci/cm}^3$, which is equivalent to 2.1×10^{-4} ppb at the containment vessel pressure of 1.57×10^{5} Pa and the assumed temperature of 323 K. This is shown in Figure 5-8 as a solid circle. As a result, the computed tritium concentration by the TPAC showed good agreement with analytical solutions and THYTAN results. However, the computed solutions were less than the reported experimental data observed in 1971 during the special tritium survey.



Figure 5-8. Comparison of tritium concentrations in the containment vessel of the Peach Bottom reactor.

5.4 Verification of Purification System Model

Verification of purification model in the TPAC is described in this section. Figure 5-9 shows the TPAC modeling for verification.



Figure 5-9. Tritium purification modeling by INL Tritium code.

The TPAC purification system model was verified by

$$V_{j} \frac{dC_{i,j}}{dt} = -R_{PF,i,j}$$
$$= -F_{PF,He} \cdot \eta_{i} \cdot C_{i,j} .$$
(5-29)

The solution for $C_{i,i}$ is found in

$$C_{i,j}(t) = C_{0,i,j} \cdot \exp\left(-\frac{F_{PF,He} \cdot \eta_i}{V_j} \cdot t\right)$$
(5-30)

where

 C_{0ii} = initial concentration of chemical i in node j [m³ (STP)/m³ (STP)].

Table 5-12 summarizes the boundary conditions for verification of the purification system model. The HT concentration was calculated during 3,000 sec.

Table 5-12. THYTAN boundary conditions for verification of the purification system model (Ohashi and Sherman 2007).

Parameter	Symbol	Unit	Value
Inventory	V_{j}	m ³ (STP)	1
Helium flow rate at purification system	$F_{PF,He}$	m ³ (STP)	1.0×10^{-3}
Fractional efficiency of purification system for HT	η_{HT}		0.9
Initial concentration of HT	$C_{0,HT,j}$	m^3 (STP)/ m^3 (STP)	1.0×10^{-9}

Figure 5-10 shows the comparisons between analytical solution and computational results by the TPAC. As shown in this figure, both the analytical solution and the TPAC solutions are in good agreement.

5.5 Verification of Total System Model

In this part, verification of the TPAC has been carried out against Peach Bottom reactor Core 2 operation. The simplified flow sheet of the reactor primary circuit and cleanup systems of the Peach Bottom HTGR are shown in Figure 5-11. A small portion of the tritium picked up by the fuel element purge flow (i.e., 454 kg/h [1,000 lb/h]) was permanently retained in the fission product traps that are located within the base of each fuel element. However, a major portion of the tritium left the core in the purge stream and entered the fission product trapping system. It was concluded that the water and Freon-cooled delay beds in the fission product trapping system did not play a significant role in the disposition of tritium in the reactor. About 10% of the purified helium leaving the Freon-cooled bed was diverted to a liquid nitrogen trap and (liquid nitrogen-cooled charcoal beds). It was confirmed that tritium was effectively captured in these traps. A smaller helium purge flow (i.e., 91 kg/h [200 lb/h]) was drawn from the steam generator to the chemical cleanup system, which consisted of an oxidizer unit and a molecular-sieve dehydrator. Gaseous HT was converted to HTO in the oxidizer and drawn off the waste disposal drain tank from the condenser downstream from the unit. Additional HTO was removed by the molecular-sieve dehydrator downstream from the oxidizer. Purified helium that exited from the chemical cleanup system upstream from the liquid nitrogen trap.



Figure 5-10. Comparisons between analytical solution and TPAC solutions for verification of the purification system.



Figure 5-11. Simplified flow sheet of the Peach Bottom HTGR, illustrating tritium transport paths (Wichner and Dyer 1979).

Figure 5-12 shows the TPAC modeling for the Peach Bottom reactor. The system is composed of reactor, steam generator, concentric pipe, containment, and purification system. For simplicity, two steam generator loops have been simplified to a single steam generator loop in the modeling. The tritium flow path in this system is as follows: The tritium is generated in the reactor core and released; majority of the released tritium enters into the main flow distributed in the whole system; the rest of the tritium from core is purged to the purification system for removing tritium from the primary loop; some of the tritium in the main flow is permeated to secondary side through the steam generator walls or leaked to the containment through the pipe lines; and a small portion of main flow is purged to another purification system.

The total volume of the system was set to be 232 m³, and the size of the each component was assumed based on the THYTAN code input developed by Ohashi and Sherman (2007). Table 5-13 summarizes the sizes of the modeling components. The temperature and the pressure of all the components were set as 809 K and 23 atm, which are the average temperature and pressure in the system.



Figure 5-12. TPAC modeling for Peach Bottom reactor.

Table 5-13. Component sizes of the modeling (INL tritium code) for the Peach Bottom reactor.

	Volume
Component Name	[m ³]
Reactor	10.5
Concentric Inner	10.5
Concentric Outer	10.5
S/G	158.2
Mixer	5.3
Mixer 1	10.5
Mixer 2	21.2
Mixer 3	5.3
Total System	232.0

The tritium birth rate of the core was determined between 310 Ci $(1.15 \times 10^{13} \text{ Bq})$ as a minimum and 1,203 Ci $(4.45 \times 10^{13} \text{ Bq})$ as a maximum for 1,550 days of Core 2 operation. The tritium birth rate, 310 Ci is the leakage to the fission product trapping system via the fuel element purge flow, and the value of 1,203 Ci is the maximum available release rate of tritium from the core. The tritium birth rate was assumed to be constant during the operating time.

The steam generator was modeled to be a single loop by merging two steam generators into one. The steam generator model consisted of three components: super heater, evaporator, and economizer. Each component was modeled by five nodes. Figure 5-13 shows the submodel of the steam generator. The metal temperatures of the steam generator were determined based on the reactor operating conditions (see Table 5-14). The parameters for the permeation models were chosen from the Equations (5-26), (5-27), and (5-28).



Figure 5-13. Steam generator submodel of the INL tritium code for Peach Bottom reactor.

				Metal Temperature (K)	
Section	Surface Area (m ²)	Outer Radius (m)	Inner Radius (m)	Helium Side Outlet	Helium Side Inlet
Economizer	212	6.350 × 10-3	4.775 × 10-3	513	593
Evaporator	474	9.525 × 10-3	7.305 × 10-3	593	618
Super-heater	275	9.525 × 10-3	6.350 × 10-3	693	853
a. For two steam g	generators.				

Table 5-14. Dimensions of the steam generator for the Peach Bottom reactor	(Ohashi and Sherman 2007)
--	---------------------------

The concentric pipe has been modeled by two pipe components of the TPAC. To model the permeation through the concentric pipe walls, the inner pipe permeation-out port was linked to the outer pipe permeation-in port. Table 5-15 shows the dimensions of the concentric pipes. The pipe wall temperature was assumed to be 616 K, and the materials were assumed to be carbon-silicon steel, A-212 (General Atomics 1978).

Table 5-15. Characteristics of the concentric duct for the Peach Bottom reactor (Ohashi and Sherman 2007).

Parameter	Value
Length (m)	7.65a
Outer diameter of inner pipe (m)	0.8064
Inner diameter of inner pipe (m)	0.7430b

a. Daseu oli 23 it i ili.

b. Based on thickness of 3.17 cm.

The two purification systems considered in the modeling were nitrogen trap and chemical cleanup system. The efficiencies for the systems were assumed to be 100% (Wichner and Dyer 1979). The mass flow-rates for the nitrogen trap and chemical clean-up system were set as 45.4 and 90.7 kg/h, respectively. The leakage was modeled at the outer pipe of the concentric pipe, so all the tritium was assumed to be leaked to the containment through the concentric pipe. The parameters used for the leakage model are determined based on the Table 5-11 in the previous section.

Tritium concentration in the primary coolant, based on the activity of 310 Ci to the purge loop and 0 Ci to the main flow channel (case of minimum release rate) was calculated by the TPAC to be $8.91 \times 10^{-12} \text{ m}^3/\text{m}^3$, which corresponds to $8.88 \times 10^{-5} \,\mu\text{Ci/cm}^3$ based on the average temperature and pressure of the primary coolant. As for the case based on the tritium activity of 310 Ci to the purge flow and 893 Ci to the main flow channel (case of maximum release rate), tritium concentration in the primary coolant was calculated to be $3.46 \times 10^{-11} \,\text{m}^3/\text{m}^3$ ($3.44 \times 10^{-4} \,\mu\text{Ci/cm}^3$). The above results are all in good agreement with the THYTAN solutions, which considered two steam generator loops and two core regions. The reported THYTAN results are $8.16 \times 10^{-5} \,\mu\text{Ci/cm}^3$ for the minimum, and $3.38 \times 10^{-4} \,\mu$ Ci/cm³ for the maximum based on the average temperature and pressure of the primary coolant.

Figure 5-14 compares the reported experimental results (Wichner and Dyer 1979), THYTAN results (Ohashi and Sherman 2007), and the TPAC. Reported experimental data on tritium in the primary coolant were obtained in 1974 by Wichner and Dyer (1979). Tritium activity in the primary coolant (main loop HT and HTO) fluctuated in a range from about $2 \times 10^{-6} \,\mu\text{Ci/cm}^3$ to about $3 \times 10^{-4} \,\mu\text{Ci/cm}^3$ during operation in 1974. Figure 5-14 shows that the computed solutions are in the range of the experimental data. However, the computed solutions are located at the higher ranges of experimental data. It can be concluded that the computed solution from the codes (THYTAN and TPAC) is slightly conservative; however, the codes could calculate tritium concentrations at the same level as the experimental results of the Peach Bottom reactor. The calculated TPAC solutions showed good agreement with the THYTAN results, even though they used different nodalizations.



Figure 5-14. Comparisons of tritium concentration, in the Peach Bottom reactor, between observed concentration (Wichner and Dyer 1979) and the computed solutions.

6. ANALYSES ON TRITIUM BEHAVIORS IN VHTR/HTSE SYSTEM

6.1 Introduction

In the VHTR hydrogen production system, tritium permeation from the core into the produced hydrogen is a significant issue because hydrogen isotopes can easily permeate through the high temperature heat exchanger tubes and contaminate product hydrogen. If the hydrogen produced in this system is used as a product, that tritium could potentially expose individuals or be released to the environment, depending on the application. Because of this potential, tritium contamination of hydrogen was considered in this study.

Figure 6-1 illustrates a schematic of the indirect parallel system configuration between a VHTR and a HTSE system, which has been proposed by Oh et al. (2007) from their system optimization studies. In this configuration, the flow in the secondary coolant system is divided, with most of the flow going towards the PCU and the remainder going through a secondary heat exchanger (SHX) that directs heat towards the HTSE plant. The flow through the hot side of the SHX is then mixed with the flow from the PCU to feed the cold side of the IHX. However, some of the flow is diverted away from the PCU, which acts to decrease the efficiency of the cycle. There are three coolant loops. The primary coolant system contains the nuclear reactor, the hot side of the IHX, and a compressor. The secondary coolant system contains the cold side of the IHX, the hot side of the SHX, the PCU, and connecting piping, which is assumed to be short. The intermediate heat transport loop connects the secondary coolant system to the HTSE plant through several PHXs.



Figure 6-1. Indirect parallel configuration of VHTR/HTSE system (VHTR part).

For electrolysis, the steam is heated to about 800°C by the heat from SHX. The heated steam is converted into hydrogen and oxygen in the electrolyzer and discharged through the fuel and oxidizer outlet, respectively as shown in Figure 6-2. The heat of the discharged gases is recovered through three recuperators. Finally, the product gas in the fuel side contains hydrogen and steam, and the oxidizer outlet gas contains oxygen and steam. As shown in Figure 6-2, the discharged fuel steam is recycled to the inlet fuel stream, and the hydrogen gas is separated and collected in the separator. In the oxidizer outlet stream, heat is first recuperated and then the stream is run through an expander to recover work. The oxygen and water components of the stream are then separated.



Figure 6-2. Indirect parallel configuration of VHTR/HTSE system (HTSE part).

6.2 TPAC Modeling

The VHTR/HTSE system described in Figure 6-1 and Figure 6-2 has been modeled by the TPAC. Figure 6-3 shows the TPAC model for this system. In this modeling, the system has been divided into five separate loops; (1) primary side, (2) secondary side (PCU), (3) intermediate heat transfer loop, (4) HTSE (water supplier), and (5) HTSE (purge flow). The primary side model is composed of the five component blocks:

- Reactor (modeled by a TPAC reactor component block)
- Hot-leg (modeled by a TPAC VHTR pipe component block)
- Cold-leg (modeled by a TPAC VHTR pipe component block)
- IHX (modeled by a TPAC heat exchanger block)
- Purification system (modeled by a TPAC purification system component block)

The secondary side (PCU) was modeled by the following seven component blocks:

- PCU (modeled by a TPAC VHTR pipe component block)
- Bypass (modeled by a TPAC VHTR pipe component block)
- IHX (modeled by a TPAC heat exchanger block)
- SHX (modeled by a TPAC heat exchanger block)
- Flow Splitter (modeled by a TPAC flow splitter component block)
- Flow Mixer (modeled by a TPAC flow mixer component block)
- Purification system1 (modeled by a TPAC purification system component block)

The intermediate heat transfer loop was modeled by the following eight component blocks:

- IHTL (modeled by a TPAC VHTR pipe component block)
- SHX1 (modeled by a TPAC heat exchanger block)
- PHX-mod (modeled by a modified TPAC heat exchanger block)
- PHX-mod1 (modeled by a modified TPAC heat exchanger block)
- PHX-mod3 (modeled by a modified TPAC heat exchanger block)
- Flow Splitter1 (modeled by a TPAC flow splitter component block)
- Flow Mixer1 (modeled by a TPAC flow mixer component block)
- Purification system2 (modeled by a TPAC purification system component block)

The HTSE system has been simplified from the original system shown in Figure 6.2. In the TPAC model, recycling and recuperating cycles have been eliminated. This simplification is expected to be more conservative than the original configuration, providing more HT permeation from the PHXs. In the TPAC model, the HTSE (water supplier) consists of the following seven component blocks:

- H₂O Flow (modeled by a TPAC boundary component block)
- Pipe-HTSE (modeled by a TPAC HTSE pipe component block)
- Pipe-HTSE1 (modeled by a TPAC HTSE pipe component block)
- Pipe-HTSE2 (modeled by a TPAC HTSE pipe component block)
- PHX-mod (modeled by a modified TPAC heat exchanger block)
- PHX-mod1 (modeled by a modified TPAC heat exchanger block)
- Isotope Exchange (modeled by a TPAC isotope exchange component block)
- Electrolyzer (modeled by a TPAC electrolyzer component block)

The HTSE (purge flow) was modeled by the following four component blocks:

- Purge Flow (modeled by a TPAC boundary component block)
- Pipe-HTSE3 (modeled by a TPAC HTSE pipe component block)
- Electrolyzer (modeled by a TPAC electrolyzer component block)
- PHX-mod3 (modeled by a modified TPAC heat exchanger block)



Figure 6-3. TPAC modeling for VHTR/HTSE system.

6.2.1 Tritium Source

Tritium sources in the reactor system are the most important, but quantitatively, very uncertain parameters in tritium modeling. The tritium sources are significantly affected by impurities (⁶Li, ⁷Li ¹⁰B, ³He ...), neutron flux, reactor power, adsorption, etc. In this modeling, these parameters have been determined based on the previous HTGR experience. The reported calculation results on tritium birth rates are well summarized by Ohashi and Sherman (2007) for various reactors: Peach Bottom (USA), Fort St. Vrain (USA), PNP-500 MWth (Germany), and England 1,500 MWth HTGR (UK). Tables 6-1 and 6-2 summarize the computed results. According to these data, the tritium birth rate per unit power (MWth) ranges from 2.14 x 10¹¹ to 4.28 x 10¹¹ Bq/y/MWth.

Countr	у	United States			Germany		England	
Reactor T	Type Peach Bottom ^a Fort St. Vrain ^b		3,000 MWth HTGR	PNP-500 MWt-plant ^b		1,500 MWth HTGR		
Reactor power (MWth)	-	66.5 ^c	842 ^d		3,000	500		1,500
Period		1,550 days	First year	Sixth year	First year	First year	Equilibrium	After 5 or 6 years
Service factor	(%)	100	80		80	100	100	90
Annual birth rate (Bq/y)	Fission	1.05 × 1013 (54.7)	1.15 × 1014 (55.3)	1.01 × 1014 (69.8)	4.07 × 1014 (59.7)	8.88 × 1013 (46.3)	1.18 × 1014 (84.0)	3.68 × 1014 (63.4)
	6Li	5.58 × 1011 (2.9)	5.33 × 1013 (25.6)	1.78 × 1013 (12.3)	1.04 × 1014 (15.3)	3.85 × 1013 (20.1)	5.55 × 1012 (3.9)	4.40 × 1013 (7.6)
	3He	5.43 × 1011 (39.6)	2.37 × 1013 (11.4)	1.37 × 1013 (9.5)	1.36 × 1014 (20.0)	5.48 × 1013 (28.6)	7.40 × 1013 (5.2)	1.68 × 1014 (29.1)
	10B	7.63 × 1012 (2.8)	1.61 × 1013 (7.7)	1.22 × 1013 (8.4)	3.40 × 1013 (5.0)	9.62 × 1012 (5.0)	9.62 × 1012 (6.8)	e
	Total	1.93 × 1013	2.09 imes 1014	1.44×1014	6.81 imes 1014	1.92×1014	1.41×1014	5.80 × 1014
Annual birth	Fission	1.59 imes 1011	1.71×1011	1.50×1011	1.70 imes 1011	1.78 imes 1011	2.37 imes 1011	2.71 × 1011
rate per	6Li	8.39 imes 109	7.92 imes 1010	2.64×1010	4.34×1010	7.70 imes 1010	1.11 imes 1010	3.24 × 1010
thermal megawatt ^f	3He	8.16 imes 109	3.52×1010	2.03×1010	5.67×1010	1.10 imes 1011	1.48×1010	1.24×1011
(Bq/y/MWth)	10B	2.54 × 109	2.39×1010	1.81 imes 1010	1.42×1010	1.92 imes 1010	1.92×1010	_ ^e
	Total	2.90×1011	3.10 × 1011	2.14×1011	2.84×1011	3.83 × 1011	2.82×1011	4.28×1011

Table 6-1. Calculation results for the tritium birth rate in the high-temperature gas-cooled reactor (Ohashi and Sherman 2007).

The value in parentheses is the percentage

a. annual birth rate calculated from the tritium inventory at the end of each period

b. annual birth rate calculated from the reported value of the birth rate per day (Ci./d) assuming 100% service factor

c. average power based on rated power of 115 MWth and equivalent full-power days of 897 days during 1,550 days

d. based on calculation condition in the reference, whereas the rated power of the Fort St. Vrain is 760 MWth

e. not estimated

f. at service factor of 100%

HTGR = high-temperature gas-cooled reactor.

Country		United States		Germany	England
			3,000 MWth		1,500 MWth
Reactor Type	Peach Bottom	Fort St. Vrain	HTGR	PNP-500	HTGR
Fission yield (n/cm ² /s)	1.00×10^{-4}	1.00×10^{-4}	1.00×10^{-4}	1.30×10^{-4}	1.55×10^{-4}
Li content in graphite (ppb)	7	50	50	10	60
³ He content in makeup helium (ppm)	0.16	0.2	0.2	0.137	0.3
Primary helium inventory (kg)	269	2,801			4,500
Helium inventory ratio in core (%)	3.25	4 to 6	6		20
Helium leak rate (1/d)	$9.95\times 10^{\text{-3a}}$	2.74×10^{-5}	_	1×10^{-4}	2.0×10^{-3}
Thermal neutron flux (n/cm ² /s)	2.28×10^{13}	6.00×10^{13}	1×10^{-14}		2.8×10^{13}
Fast neutron flux (n/cm ² /s)	2.26×10^{13}	3.60×10^{13}			
Cross section for ⁶ Li (n, α) ³ H (cm ²)	4.08×10^{-22}	$4.08\times10^{\text{-}22}$	4.63×10^{-22}		$9.30\times10^{\text{-}22}$
Cross section for ${}^{3}\text{He}(n, p){}^{3}\text{H}(cm^{2})$	2.28×10^{-21}	2.28×10^{-21}	2.80×10^{-21}		5.40×10^{-21}
Cross section for ${}^{10}B(n, \alpha)^7Li(cm^2)$	1.63×10^{-21}	1.63×10^{-21}	—		
Cross section for ⁷ Li (n, $n\alpha$) ³ H (cm ²)	1.53×10^{-25}	7.2×10^{-27}			
Cross section for ${}^{10}B(n, 2\alpha){}^{3}H(cm^{2})$	5.0×10^{-26}	1.4×10^{-26}			
Reference	Wichner and	Compare,	Gainey 1976	Steinwarz,	Forsyth 1974
	Dyer 1979	Freid, and		Rohrig, and	
		Nestor 1974		Nieder 198	
a. Based on leak rate of 3.2 kg/d and primary helium inventory of 232 m3 (average temperature of 809 K and pressure of 23 atm).					

Table 6-2. Additional calculation results for the tritium birth rate in the high-temperature gas-cooled reactor (Ohashi and Sherman 2007).

a. Based on leak rate of 3.2 kg/d and primary helium inventory of 232 m3 (average temperature of 809 K and pressure of 23 atm). HTGR = high-temperature gas-cooled reactor

In our modeling, the tritium birth rate, 4.28×10^{11} Bq/y/MWth, was selected in a conservative manner. The tritium release rate was determined based on the previous investigation of the tritium inventory in the core components of the Peach Bottom HTGR performed by Wichner and Dyer (1979). According to their studies, the overall release fraction to the primary coolant is estimated to be 0.32 during 3 years of operation (Burnette and Baldwin 1980). In our modeling, this number (~0.32) was assumed as the base fraction of tritium release rate. However, since the tritium retention and release mechanisms are still not well understood, the assumed tritium release ratio is basically considered to have large uncertainties. The effect of the tritium release uncertainty on the tritium source (release rate) in our model was finally determined as 8.28×10^{13} Bq/y (=5.49 $\times 10^{-11}$ m³ (STP)/s) for 600 MWth reactor power. This number was implemented to the 'Tritium Source' block attached to the 'Reactor block' shown in Figure 6-3.

6.2.2 Component Inputs

This section summarizes input parameters of the component blocks used in the TPAC modeling.

6.2.2.1 Component Input Summary in Primary Side

This section summarizes input parameters of the component blocks that consist of the primary side. The parameters related to the system operating conditions (temperature, pressure, and flow rates) are taken from the flow sheet shown in Figure 6-1. The geometrical information (volume, wall thickness, etc.) has been estimated based on the GTMHR preconceptual designs and many assumptions. However, the geometrical parameters basically do not affect the calculation results assuming the steady-state

conditions. However, the TPAC has been originally developed for dynamic simulations, so it is also capable of flexible dynamic simulations.

Tables 6-3 through 6-8 summarize the input parameters implemented in the primary loop model. The tritium release rate is set to be 5.49×10^{-11} m³(STP)/s in the 'Tritium Birth' block as already mentioned in the previous section. Initial HT and H₂O concentrations in the components were assumed to be 1.0×10^{-20} and 1.0×10^{-5} in mole fraction, respectively. Since the current simulation is under steady state conditions, the initial concentrations in the components do not affect the final results. Helium leakage was considered in the 'Reactor' block. Fractional helium leak rate was assumed to be 1.15×10^{-7} , which is the same as that reported for Peach Bottom reactor in Section 5. The IHX design parameters (surface area, wall thickness, etc.) are based on the PCHE design reported by Oh and Kim (2008). The permeation parameters (activation energy, and pre-exponential factor) were based on the Incoloy 800 material data reported by Richards et al. (2006). Permeation of H₂ was not considered in the current simulation. Therefore, the pre-exponential factor for H₂ permeation was set to be 0.0. The purification rate was determined to be 16%/h, which is the average rate between Peach Bottom HTGR (20%/h) and Fort St. Vrain (12%/h). Purification efficiency for HT was assumed to be 95% conservatively. The details of the input parameters can be referred to Tables 6-3 through 6-8.

Parameter	Unit	Value
Tritium Release Rate	Bq/y	8.28E+13
	m ³ (STP)/s	5.49E-11

 Table 6-3. TPAC input parameters in the primary side (Tritium Birth Block).

 Tritium Dirth

1 able 6-4.	IPAC	input	parameters	in the	primary	side	(Reactor	Block).

Reactor		
Parameter	Unit	Value
Volume	m ³	197
Temperature	K	1.05E+03
Pressure	Ра	7.00E+06
Flow Rate	m ³ (STP)/s	2,155
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	YES
Total Inventory in Loop	m ³ (STP)	4,772
Fractional Helium Leak Rate	1/s	1.15E-7
Sum of Inventory with Helium Leak	m ³ (STP)	4,772
Permeation	#	NO

1101 205		
Parameter	Unit	Value
Volume	m ³	12
Temperature	K	1.17E+03
Pressure	Ра	7.00E+06
Flow Rate	m ³ (STP)/s	2155
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO

 Table 6-5. TPAC input parameters in the primary side (Hot Leg Block).

 Hot Leg

Table 6-6. TPAC input parameters in the primary side (Cold Leg Block).

Cold Leg		
Parameter	Unit	Value
Volume	m ³	19
Temperature	Κ	9.31E+02
Pressure	Ра	7.00E+06
Flow Rate	m ³ (STP)/s	2,155
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO

Parameter	Unit	Value	
Volume	m ³	13	
Temperature	К	1.05E+03	
Pressure	Ра	7.00E+06	
Flow Rate	m ³ (STP)/s	2,155	
C _{HT} Initial	mf	1.00E-20	Channel I
C _{H2} Initial	mf	1.00E-05	
Leakage	#	NO	
Permeation	#	NO	
Volume	m ³	13	
Temperature	К	1.05E+03	
Pressure	Ра	7.00E+06	
Flow Rate	m ³ (STP)/s	2,155	Channal 2
C _{HT} Initial	mf	1.00E-20	Channel 2
C _{H2} Initial	mf	1.00E-05	
Leakage	#	NO	
Permeation	#	NO	
Heat Transfer Area	m ²	1.35E+04	
Wall Thickness	m	0.02	
HX Wall Temperature	К	1.06E+03	
Pre-exponential Factor (HT)	-	1.36E-10	Channel Wall
Pre-exponential Factor (H ₂)	-	0	
Activation Energy (HT)	J/mol K	5.35E+04	
Activation Energy (H ₂)	J/mol K	NONE	

 Table 6-7. TPAC input parameters in the primary side (IHX Block).

 IHX

Table 6-8. TPAC input parameters in the primary side (Purification System Block).

Purification System		
Parameter	Unit	Value
Fractional Mass Flow to Purification System		9.84E-05
HT Purification Efficiency	(0 ~ 1)	0.95
H ₂ Purification Efficiency	(0 ~ 1)	0
H ₂ O Purification Efficiency	(0 ~ 1)	0
HTO Purification Efficiency	(0 ~ 1)	0
6.2.2.2 Component Input Summary in Secondary Side (PCU)

This section summarizes input parameters of the component blocks used in the secondary side (PCU) modeling. The parameters related to the system operating conditions (temperature, pressure, and flow rates) are taken from the flow sheet shown in Figure 6-1. Tables 6-9 through 6-13 summarize the input parameters implemented in the secondary side model.

100		
Parameter	Unit	Value
Volume	m ³	197
Temperature	Κ	1.02E+03
Pressure	Ра	5.41E+06
Flow Rate	m ³ (STP)/s	1,679
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	YES
Total Inventory in Loop	m ³ (STP)	3,682
Fractional Helium Leak Rate	1/s	1.15E-7
Sum of Inventory with Helium Leak	m ³ (STP)	3,682
Permeation	#	NO

Table 6-9. TPAC input parameters in the secondary side (PCU Block).

Fable 6-10. TPAC inpu	parameters in the s	secondary side	(Purification	System Block).
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Purification System1		
Parameter	Unit	Value
Fractional Mass Flow to Purification System	—	9.75E-05
HT Purification Efficiency		0.95
H ₂ Purification Efficiency		0
H ₂ O Purification Efficiency		0
HTO Purification Efficiency		0

Table 6-11. TPAC inp	ut parameters in the sec	ondary side (Bypass Block).
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Bypass		
Parameter	Unit	Value
Volume	m ³	31
Temperature	K	1.16E+03
Pressure	Ра	7.54E+06
Flow Rate	m ³ (STP)/s	141.036
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO

SHX		
Parameter	Unit	Value
Volume	m ³	1
Temperature	K	9.82E+02
Pressure	Ра	7.54E+06
Flow Rate	m ³ (STP)/s	141.036
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO
Volume	m ³	1
Temperature	K	9.71E+02
Pressure	Ра	5.00E+06
Flow Rate	m ³ (STP)/s	142.23
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO
Heat Transfer Area	m ²	1,519
Wall Thickness	m	0.028
HX Wall Temperature	K	999
Pre-exponential Factor (HT)	-	1.36E-10
Pre-exponential Factor (H ₂)	-	0
Activation Energy (HT)	J/mol K	5.35E+04
Activation Energy (H ₂)	J/mol K	NONE

Table 6-12. TPAC input parameters in the secondary side (SHX Block).

Table 6-13. TPAC input parameters in the secondary side (Splitter Block).

Parameter	Unit	Value
Separation Ratio		0.066

6.2.2.3 Component Input Summary in Intermediate Heat Transfer Loop

This section summarizes input parameters of the component blocks used in the intermediate heat transfer loop (IHTL) modeling. The parameters related to the system operating conditions (temperature, pressure, and flow rates) are taken from the flow sheet shown in Figure 6-1. Tables 6-14 through 6-19 summarize the input parameters implemented in the intermediate heat transfer loop model. In the ITL modeling, the PHXs were designed to permeate tritium in one side from the IHTL to the HTSE system. This simplification is based on the quite reasonable assumption that there is no tritium in the inlet water flow in the HTSE system. This simple modification to the heat exchanger model made a large enhancement to the computational speed.

Parameter	Unit	Value
Volume	m ³	40
Temperature	К	7.47E+02
Pressure	Ра	2.00E+06
Flow Rate	m ³ (STP)/s	106
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO

Table 6-14. TPAC input parameters in the IHTL side (IHTL Block). IHTL

 Table 6-15. TPAC input parameters in the IHTL side (PHX-mod Block).

 PHX-mod

	•	
Parameter	Unit	Value
Volume	m ³	1
Temperature	К	7.53E+02
Pressure	Pa	2.00E+06
Flow Rate	m ³ (STP)/s	53
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO
Volume	m ³	12
Temperature	K	7.53E+02
Pressure	Ра	5.00E+06
Flow Rate	m ³ (STP)/s	0.1
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO
Heat Transfer Area	m ²	34
Wall Thickness	m	0.01
HX Wall Temperature	Κ	753
Pre-exponential Factor (HT)	-	1.36E-10
Pre-exponential Factor (H2)	-	0
Activation Energy (HT)	J/mol K	5.35E+04
Activation Energy (H2)	J/mol K	NONE

111/1-111001		
Parameter	Unit	Value
Volume	m ³	1
Temperature	K	9.53E+02
Pressure	Ра	2.00E+06
Flow Rate	m ³ (STP)/s	53
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO
Volume	m3	1
Temperature	K	9.53E+02
Pressure	Ра	2.00E+06
Flow Rate	m ³ (STP)/s	53
C _{HT} Initial	mf	1.00E-20
C _{H2} Initial	mf	1.00E-05
Leakage	#	NO
Permeation	#	NO
Heat Transfer Area	m^2	1,287
Wall Thickness	m	0.01
HX Wall Temperature	Κ	971
Pre-exponential Factor (HT)	-	1.36E-10
Pre-exponential Factor (H2)	-	0
Activation Energy (HT)	J/mol K	5.35E+04
Activation Energy (H2)	J/mol K	NONE

 Table 6-16. TPAC input parameters in the IHTL side (PHX-mod1 Block).

 PHX-mod1

Unit	Value
m ³	1
К	9.53E+02
Ра	2.00E+06
m ³ (STP)/s	53
mf	1.00E-20
mf	1.00E-05
#	NO
#	NO
m3	1
К	9.53E+02
Ра	2.00E+06
m ³ (STP)/s	53
mf	1.00E-20
mf	1.00E-05
#	NO
#	NO
m^2	2,161
m	0.01
К	971
-	1.36E-10
-	0
J/mol K	5.35E+04
J/mol K	NONE
	Unit m ³ K Pa m ³ (STP)/s mf mf # # m3 K Pa m ³ (STP)/s mf mf # # # mf mf K Pa m ³ (STP)/s Mf mf Mf mf - - J/mol K J/mol K

Table 6-17. TPAC input parameters in the IHTL side (PHX-mod2 Block). PHX-mod2

Table 6-18. TPAC input parameters in the IHTL side (Flow Splitter1 Block).

Flow Splitter1		
Parameter	Unit	Value
Separation Ratio	-	0.5

 Table 6-19. TPAC input parameters in the IHTL side (Purification System2 Block).

 Purification System2

Parameter	Unit	Value
Fractional Mass Flow to PS	-	1.41E-04
HT Purification Efficiency	-	0.95
H2 Purification Efficiency	-	0
H2O Purification Efficiency	-	0
HTO Purification Efficiency	-	0

6.2.2.4 Component Input Summary in HTSE

This section summarizes input parameters of the component blocks used in the HTSE loop. The parameters related to the system operating conditions (i.e. temperature, pressure and flow rates) have been taken from the flow sheet shown in Figure 6-1. Tables 6-20 through 6-26 summarize the input parameters implemented in our HTSE model.

HZO FIOW		
Parameter	Unit	Value
Flow Rate	m ³	18.8
Temperature	К	3.0E+02
Pressure	Ра	101300
C _{HT}	mf	0
C _{H2}	mf	0
C _{H2O}	mf	1
С _{нто}	mf	0
C _{H2SO4}	mf	0
C _{HTSO4}	mf	0
C _{HI}	mf	0
C _{TI}	mf	0

 Table 6-20. TPAC input parameters in the HTSE components.

 H2O Flow

Table 6-21. TPAC input	parameters in	the HTSE components ((Pipe-HTSE Block).
		_	

Pipe-HTSE		
Parameter	Unit	Value
Volume	m ³	10
Temperature	K	3.0E+02
Pressure	Ра	5.00E+06
Flow Rate	m ³ (STP)/s	18.8
C _{HT} Initial	mf	0
C _{H2} Initial	mf	0
C _{H2O} Initial	mf	1
C _{HTO} Initial	mf	0
Leakage	#	NO
Permeation	#	NO

Parameter	Unit	Value
Volume	m ³	10
Temperature	K	3.0E+02
Pressure	Ра	5.00E+06
Flow Rate	m ³ (STP)/s	18.8
C _{HT} Initial	mf	0
C _{H2} Initial	mf	0
C _{H2O} Initial	mf	1
C _{HTO} Initial	mf	0
Leakage	#	NO
Permeation	#	NO

 Table 6-22. TPAC input parameters in the HTSE components (Pipe-HTSE1 Block).

 Pipe-HTSE1

Table 6-23. TPAC in	put parameters ir	n the HTSE components ((Pipe-HTSE2 Block).
		_	

Pipe-HTSE2		
Parameter	Unit	Value
Volume	m ³	10
Temperature	Κ	6.40E+02
Pressure	Ра	5.00E+06
Flow Rate	m ³ (STP)/s	18.8
C _{HT} Initial	mf	0
C _{H2} Initial	mf	0
C _{H2O} Initial	mf	1
C _{HTO} Initial	mf	0
Leakage	#	NO
Permeation	#	NO

Table 6-24. TPA	C input	parameters	in the HTSE	components	(Pipe-HTSE3	Block).
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Pipe-HTSE3		
Parameter	Unit	Value
Volume	m ³	10
Temperature	Κ	3.0E+02
Pressure	Ра	5.00E+06
Flow Rate	m ³ (STP)/s	11.2
C _{HT} Initial	mf	0
C _{H2} Initial	mf	0
C _{H2O} Initial	mf	1
C _{HTO} Initial	mf	0
Leakage	#	NO
Permeation	#	NO

Table 6-25. TPAC input parameters in the HTSE components (Electrolyzer Block). Pine-HTSE3

Parameter	Unit	Value
Steam Conversion Ratio	%	90

Table 6-26. TPAC input parameters in the HTSE components (Purge Flow Block).

1120110₩		
Parameter	Unit	Value
Flow Rate	m ³	11.2
Temperature	К	3.0E+02
Pressure	Ра	101,300
C _{HT}	mf	0
C _{H2}	mf	0
C _{H2O}	mf	1
С _{нто}	mf	0
C _{H2SO4}	mf	0
C _{HTSO4}	mf	0
C _{HI}	mf	0
C _{TI}	mf	0

6.2.3 Simulation and Results of the Base Case

This section summarizes the results of tritium analyses based on the reference configuration and data described in the previous sections. Figure 6-4 shows the solver setting options used for our analyses. In the current simulation, the ode15s solver was used while the default solver in the MATLAB SIMULINK is ode45, which is based on an explicit Runge-Kutta (4,5) formula. This is because, in general, ode45 is the best solver to apply as a first try for most problems. However, for a stiff problem, ode45 is inefficient because solutions can change on a time scale that is very short compared to the interval of integration and the ode45 solver has not been designed for this situation. The ode15s solver is a variable-order solver based on the NDFs (Mathworks 2009). These are related to but are more efficient than the backward differentiation formulas (BDFs) (also known as Gear's method). This simulation is suspected to be a stiff problem because of the large flow rate in the system. Ode15s therefore seems more adequate than ode45. In the internal benchmark calculations, the ode15s solver showed significantly faster computational speed than the ode45 solver especially for this input model. Maximum time step, Minimum time step, and Initial time step were automatically decided by the MATLAB code. The simulation time was 5 x 10^5 sec (~5.79 days).

elect:	Simulation time					
Solver Data Import/Export	Start time: 0.0			Stop time: 5e5		
Optimization]Diagnostics	Solver options					
Sample Time	Type:	Variable-step	T	Solver:	ode15s (stiff/NDF)	•
····Conversion	Max step size:	auto		Relative tolerance:	1e-3	
····Connectivity ····Compatibility	Initial step size:	auto		Maximum order:	5	•
Model Referencing	Zero crossing control:	Use local settings	•		, -	_
Comments Symbols Custom Code Debug Interface						

Figure 6-4. TPAC solver setting options for tritium analyses of the VHTR/HTSE system.

Figures 6-5 through 6-8 show time variations of the HT concentrations in the primary side, secondary side, intermediate heat transfer loop, and product hydrogen, respectively. These figures clearly show that the HT concentrations have reached steady-state successfully. In these simulations, steady-state has been achieved in about 2×10^5 sec after starting.



Figure 6-5. HT concentrations in the primary side (Reactor block).



Figure 6-6. HT concentrations in the secondary side (PCU block).



Figure 6-7. HT concentrations in the IHTL side (IHTL block).



Figure 6-8. HT concentrations in product hydrogen.

Figure 6-9 shows the tritium distribution computed by the TPAC for the VHTR/HTSE system. As shown, about 69.97% of the released tritium from the core is removed by the purification system installed in the primary side. About 29.99% of the tritium is transferred to the secondary side through the heat exchanger wall. Only 0.14% of tritium is released to the containment (or confinement) by helium leakage. In the secondary side, most of the tritium (28.3%) is removed by the purification system. Only 1.63% of the tritium released from the core is transferred to the intermediate heat transfer loop through the heat exchanger. In the intermediate loop, 0.56% of the tritium released from the core is removed by the purification system. About 0.5% of the tritium released from the core is transferred to the product hydrogen. According to the calculation results, the purification systems in the primary and secondary system are the most in need of tritium transfer removal and distribution. The effect of leakage from the system looks negligible, which is less than 0.2% in total. Table 6-27 summarizes the tritium concentration in gaseous hydrogen (5.65 x 10^{-4} Bq/cm³ [STP]) is an order less than the gaseous effluent limit (3.7 x 10^{-3} Bq/cm³ [STP]).

As already mentioned, the most uncertain parameter based on these analyses is the tritium release rate. Some parametric studies have been performed here for looking at the effect of tritium release rate on the tritium concentration in the hydrogen product. Fractions of tritium release rate to tritium birth rate have been changed from 0.0 to 1.0 in this study. Figure 6-10 shows the computed results. In this graph, x-axis represents the ratio of tritium release to birth rate, and y-axis represents tritium concentration in the product hydrogen. Our analyses showed that the tritium concentration in the hydrogen product is linearly related to the tritium release ratio. One notable result is that even in the most serious conditions when all the tritium generated from the core is released to the reactor primary coolant (ratio of tritium release to birth rate = 1.0), the tritium concentration $(1.17 \times 10^{-3} \text{ Bq/cm}^3)$ in the product hydrogen is still much lower than that of the gaseous tritium effluent limit $(3.7 \times 10^{-3} \text{ Bq/cm}^3)$.



Figure 6-9. Tritium distribution in the VHTR/HTSE system.

Location	[ppb]	[µCi/cm ³ [STP]]	[Bq/cm ³ [STP]]	Limit
Primary	1.906 x 10 ⁻¹	2.48 x 10 ⁻⁴	9.15	
Secondary	9.973 x 10 ⁻²	1.30 x 10 ⁻⁴	4.79	_
IHTL	2.123 x 10 ⁻²	2.76 x 10 ⁻⁵	1.02	
Product Hydrogen	1.176 x 10 ⁻⁵	1.58 x 10 ⁻⁸	5.65 x 10 ⁻⁴	3.7 x 10 ⁻³

Table (27	Taitima	a an a antination	in the			. IIITI		nna dae at headn	~ ~ ~ ~
Table $0-2/$.	IIIIIIIII	concentration	in the	primary,	secondary	/ IN I L,	and	product fiyar	ogen.



Figure 6-10. Effect of tritium release ratio on the tritium concentration in the hydrogen product.

7. SUMMARY

This study focuses on one important potential issue for application of the VHTRs to hydrogen or process steam production. A serious concern for VHTRs is tritium permeation from the core into the product hydrogen or process steam. Tritium can be easily permeated through high temperature heat exchanger (or steam generator) tubes or heat transfer surfaces, thereby contaminating coolant or product gases. Tritium is mostly generated in the reactor by ternary fission of the fuel and neutron reactions with lithium impurities in the graphite, boron control materials in the reflector block, and ³He naturally occurring in the helium coolant. Tritium that enters or is formed in the helium primary coolant will be circulated or permeate to the secondary coolant and the intermediate heat transfer loop. The permeated tritium then enters the product hydrogen or process steam through contaminated heat exchanger (or steam generator) surfaces.

A computer code was developed using MATLAB SIMULINK to analyze tritium behaviors in the VHTR systems. The code was named TPAC (Tritium Permeation Analyses Code). TPAC is based on the mass and species conservation laws of the tritium or hydrogen containing molecules such as HT, H₂, HTO, H₂O, H₂SO₄, HTSO₄, TI, etc. Various tritium birth models have been implemented into the code. The models are ternary fission and neutron reactions with impurities (⁶Li, ⁷Li, ¹⁰B, ³He) in core structures and coolant. A tritium permeation model using the Arrhenius-type equation form was considered in heat exchanger, vessel, and pipe walls. The default parameters are based on the Alloy-800 materials, but the parameters can be easily modified in the input panel. A leakage model, purification system model, electrolyzer model, and isotope-exchange model are also implemented in the code. Based on these models, various system component models related to VHTRs and hydrogen production systems were developed. The developed components include reactor, vessel, pipe, containment, purification system, heat exchangers, etc. Data access and display blocks are also implemented into the code for post-processing. The basic models used in the code are well described in Chapter 3, and the code details and user-guide are documented in Chapter 4, with some tutorials.

The TPAC has been verified by analytical solutions and experimental data based on the Peach Bottom reactor. Various models were taken into consideration for verification (tritium birth models, leakage model, permeation model, purification system model, isotope-exchange model, etc.). The TPAC calculation results showed good agreement with the analytical and experimental data. The total system model for the Peach Bottom reactor was analyzed by the TPAC and compared with the reported data. The TPAC showed reasonable agreement with the reported data in the Peach Bottom reactor giving a conservative prediction. However, the quality of the measured data was not good enough for validation, and the input conditions seem to have large uncertainties. The verification works are well documented in Chapter 5.

Analyses of tritium behaviors in the VHTR/HTSE system were carried out using the TPAC. The reference reactor configurations and operating conditions were taken from the indirect parallel configuration proposed by Oh et al. (2007) in their system optimization studies. The geometrical information for the reactor has been referred to the current VHTR preconceptual designs with some assumptions. Heat exchanger designs including IHX, SHX, and PHXs were based on the PCHE-type heat exchangers and the design specifications were taken from the report published by Oh and Kim (2008). Information related to the tritium birth, purification, and leakages were based on various previous literatures (Richards 2006, General Atomics 2006, Ohashi and Sherman 2007), Wichner and Dyer 1979). This analysis showed that only 0.4% of tritium released from the core is transferred to the product hydrogen. The amount of tritium in the product hydrogen ($5.65 \times 10^{-4} \text{ Bq/cm}^3$) was estimated to be about an order of magnitude less than the gaseous effluent limit for tritium ($3.7 \times 10^{-3} \text{ Bq/cm}^3$). A parametric study for the tritium release, which has the largest uncertainties for the simulations, showed that even in the most serious conditions— when all of the tritium generated from the core is released to the reactor

primary coolant (ratio of tritium release to birth rate = 1.0)—the tritium concentration $(1.17 \times 10^{-3} \text{ Bq/cm}^3)$ in the product hydrogen is still much lower than the gaseous tritium effluent limit $(3.7 \times 10^{-3} \text{ Bq/cm}^3)$.

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Code and Version	Computer Platform	Operating System
MATLAB 7.8.0 (R2009a)	HPC Cluster (icestorm) PC (OPTIPLEX 760)	SUSE Linux Enterprise Server 10 (x86_64) VERSION = 10 PATCHLEVEL = 2 SGI ProPack 6SP2 for Linux, Build 602r4-0812292001 kernel version: 2.6.16.60-
		0.37_f594963d-smp Windows XP version 5.1 sp3
TPAC 1.5b	PC (OPTIPLEX 760) (Executable in the HPC Cluster)	Windows XP version 5.1 sp3

Computer Platform Information